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journal homepage: www.elsevier.com/locate/jqsrtEmpirical rovibrational energy levels of ammonia up to 7500 cm⁻¹Tibor Furtenbacher^a, Phillip A. Coles^b, Jonathan Tennyson^b, Sergei N. Yurchenko^b, Shanshan Yu^c, Brian Drouin^c, Roland Tóbiás^a, Attila G. Császár^{a,*}^a MTA-ELTE Complex Chemical Systems Research Group and ELTE Eötvös Loránd University, Institute of Chemistry, Laboratory of Molecular Structure and Dynamics, H-1117 Budapest Pázmány Péter sétány 1/A, Hungary^b Department of Physics and Astronomy, University College London, London WC1E 6BT, United Kingdom^c Jet Propulsion Laboratory, California Institute of Technology, Pasadena, CA 91109, USA

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ABSTRACT

The most recent version of the MARVEL (Measured Active Rotational-Vibrational Energy Levels) protocol and code is used to update and extend the list of accurately known empirical rovibrational energy levels of ¹⁴NH₃ falling in the range of 0 – 7500 cm⁻¹. Compared to an earlier similar study covering all the measured transitions of ¹⁴NH₃ [22], the present investigation is limited to transitions with an upper energy level below 7500 cm⁻¹, considers 82 sources (28 new) of transitions data, corrects several older assignments, and utilizes an improved validation procedure based partially on high-quality first-principles rovibrational energies. This study yields 4936 uniquely labelled empirical rovibrational energy levels, the set is complete up to 2413 cm⁻¹. We confirm the overall high accuracy of the most recent “spectroscopic” (fitted) ¹⁴NH₃ potential energy surface, C2018, up to 7500 cm⁻¹. Investigation of energy values resulting from previous effective-Hamiltonian fits forms an important part of this study. It is established that previous effective-Hamiltonian fits, using high expansion orders in the Hamiltonian, are inaccurate for certain high rotational excitations of all vibrational parents considered. Employing the ground vibrational state as a test case, it is shown that the origin of the problem is that the observed ¹⁴NH₃ spectroscopic network formed by rotational transitions and used for the effective-Hamiltonian fit contains floating components. A much improved effective-Hamiltonian fit is presented whereby the floating components are connected to the principal components of the spectroscopic network by carefully selected first-principles links. It is recommended that for all molecules the experimental set of transitions employed for an effective Hamiltonian fit should be checked for floating components and extra care be exercised if they exist, as in these cases the adequate reproduction of the transitions does not automatically mean that the energies derived from the effective-Hamiltonian parameters are correct as well. The connectivity problem of measured spectroscopic networks becomes more and more pronounced as the order of the Hamiltonian is increased.

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1. Introduction

Ammonia, NH₃, a simple four-atomic “inorganic” molecule, is vital to society as modern food production is highly dependent upon ammonia-based fertilizers. Ammonia is manufactured at the prodigious global rate of about 5 tonnes per second. Ammonia is a poisonous gas both for humans and plants and its ever-increasing release into the earth’s atmosphere has several undesirable consequences [1–3]. Thus, monitoring the concentration of ammonia in the atmosphere and understanding the associated nitrogen cy-

cle [2] are important scientific and engineering objectives which can be supported by detailed, high-resolution spectroscopic investigations. Measurements of spatially resolved atmospheric concentrations of ammonia require reliable and extensive laboratory data, preferably with a complete coverage of the energies and transitions of the spectroscopic regions of interest.

Astronomically, ammonia was the first polyatomic molecule detected in interstellar space [4] and, due to its thermodynamic stability, it is one of the principal molecular carriers of nitrogen. Ammonia provides one of the most accurate “molecular thermometers” for the interstellar medium [5,6] and it is omnipresent in star-forming primordial gas clouds. Ammonia is well known in solar-system gas giants and has recently been detected in the

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atmosphere of a hot-Jupiter exoplanet [7]. Ammonia is thought to provide the key signature of Y-dwarfs [8,9], the coolest category of brown-dwarf stars. Consequently, ammonia spectra have been analysed in considerable detail in brown dwarfs [10]. Detailed knowledge of the high-resolution spectrum of ammonia at widely different temperatures, from cold (10 K) to hot (2000 K), is key to the success of a large number of astronomical studies.

The rovibrational states of ammonia display characteristic and highly varying tunnelling splittings associated with ammonia's famous inversion motion [11–13]. Note that the splittings and the associated nuclear dynamics can be controlled by medium-intensity laser light [14], with possible technological applications [15]. The high-resolution rovibrational spectrum of ammonia is also relevant for metrology. Metrology studies based on precision spectroscopy and involving ammonia include the accurate determination of the Boltzmann constant [16–18] as well as the time variation of the electron-to-proton mass ratio [19–21].

Al-Derzi et al. [22] undertook a systematic compilation of a large number of measured and assigned high-resolution spectra of $^{14}\text{NH}_3$, published prior to 2014. The experimental transitions data collated by Al-Derzi et al. [22] determine 4961 empirical rovibrational energy levels. These levels were derived via the MARVEL (Measured Active Rotational-Vibrational Energy Levels) procedure [23–25], built upon the concept of spectroscopic networks [26–28]. The results of the investigation of Al-Derzi et al. [22] have been used for a number of scientific studies, including ones focused on nuclear-quadrupole coupling [29] and quadrupole splittings [30]. Since 2014, a number of new high-resolution studies of ammonia spectra have become available [30–41]. In addition, analysis of near-infrared (NIR) ammonia spectra [41] indicated that a number of transitions included in the study of Al-Derzi et al. [22] were not assigned entirely correctly, which, combined with some other issues [13], led to a number of incorrect or incorrectly assigned energy levels. A number of changes to the transitions data collated by Al-Derzi et al. [22] have already been described in Ref. [13]. For these changes the reader should consult the Supplementary Material of Ref. [13].

As part of this work, we extend and update the MARVEL analysis of the high-resolution spectra of $^{14}\text{NH}_3$. In doing so we use not only new [30–41] and re-assigned spectra but have taken the opportunity to include several older sources of spectroscopic data [42–64], which, for various reasons, were not treated by Al-Derzi et al. [22]. As a result, the current MARVEL database contains considerably more measured transitions than that employed by Al-Derzi et al. [22].

A significant portion of the recent experimental studies have concentrated on vibration-rotation spectra of ammonia with transitions above 7000 cm^{-1} [35,39–41]. Two of these investigations, Refs. [35] and [39], represent the first ever assignment of the $7400\text{--}8600$ and the $9000\text{--}11\,000\text{ cm}^{-1}$ regions. A recent study [41] found certain discrepancies between the assignments of Ref. [41] and Ref. [35]. It would therefore not be surprising if similar issues with the assignments presented in Ref. [39] appeared in the future. Until more work is done to consolidate the results of Refs. [35] and [39], it is safest not to include them in a MARVEL-based study. Thus, we restrict the present investigation to transitions lying below 7500 cm^{-1} . Al-Derzi et al. [22] considered only a single source which contained transitions above 6625 cm^{-1} . This source, 86CoLe [65], contains 323 partially assigned lines lying at about $15\,000\text{ cm}^{-1}$. These results have recently been reassessed by Zobov et al. [40] and the reader is referred to Ref. [40] for further information on observed high-resolution ammonia spectra at visible wavelengths. They will not be discussed further in this paper.

A number of variationally computed line lists are available for ammonia [40,66–70]. These can be highly useful for validating the empirical energy levels produced by the MARVEL procedure. In this

study we employ the recent C2018 [70] and CoYuTe [71] variational line lists for validation purposes. The C2018 line list, which was produced as part of the ExoMol project [72,73], is significantly more accurate than the BYTe line list [67] used for validation of the empirical rovibrational energy levels deduced by Al-Derzi et al. [22]. Overall, one of the principal aims of the present study is the production of a set of assigned rovibrational energy levels of $^{14}\text{NH}_3$ up to 7500 cm^{-1} containing as many accurate empirical levels as possible. When empirical energy levels are not available, we recommend the complete set of first-principles energy levels of Ref. [71]. Accuracy of the complete energy level set of a molecule could be improved by replacing first-principles energy values with those derived from effective Hamiltonians (EH). However, in the case of $^{14}\text{NH}_3$, as shown below, one must be particularly careful with this approach.

Historically, EHs have been considered [74–79] unsuitable for fitting the observed high-resolution spectra of ammonia to experimental accuracy. Nevertheless, recent studies by Pearson and co-workers [33,80–82] showed that carefully constructed rotation-inversion EHs are capable of describing most of the measured transitions with remarkable accuracy. The vibrational states for which rotation-inversion EHs are available include the non-interacting ground and ν_2 states, as well as the $2\nu_2 - \nu_4$ and $3\nu_2 - (\nu_2 + \nu_4)$ pairs (we employ the canonical ordering of the vibrational states throughout this paper). While it is expected that EHs may not be able to extrapolate properly, it is usually assumed that they do interpolate energy levels with an accuracy that surpasses significantly that of even the best first-principles variational computations. Thus, we decided to investigate how best to complement empirical rovibrational energies of $^{14}\text{NH}_3$ with EH ones in order to improve the accuracy of the final energy levels set of this study. The task is especially hard for high- K states for which characteristics of the ammonia spectroscopy, in particular the $\Delta K = 0$ propensity rule, make the fitting particularly challenging. As detailed in Section 3, we found significant issues with the EH fits of Refs. [33,80–82]. New EH fits, yielding much more accurate rovibrational energy values for certain highly excited rovibrational states have been made and a new set of spectroscopic parameters is provided for the vibrational ground state of $^{14}\text{NH}_3$.

2. Empirical rovibrational energy levels of $^{14}\text{NH}_3$

The MARVEL protocol [23–25], yielding empirical rovibrational energies, always starts with the construction of a spectroscopic network (SN) [26,28] for the molecule in question, here $^{14}\text{NH}_3$, using the dataset of measured and assigned transitions collated from the literature. The measured transitions must each have a *unique* assignment, which determines their place within the SN, and an uncertainty. An inversion is then performed, yielding empirically determined rovibrational energy levels within each component of the SN. Recently the algorithms employed by MARVEL have been systematically improved [23,25,83], the relationship of SNs to formal network theory considered [26,27,84], and the underlying methodology reviewed [28,85,86]. MARVEL has been used to obtain accurate empirical rovibronic energy levels for a considerable number of molecules, including diatomic [87–91], triatomic [92–101], tetraatomic [22,102], and polyatomic [103] species.

NH_3 has three identical H atoms whose interchange is therefore subject to the Pauli principle. Within the $D_{3h}(\text{M})$ molecular-symmetry (MS) group [104], the states are divided into three categories: *ortho* states have (A_2', A_2'') symmetry, *para* states have (E', E'') symmetry, and there are no states of (A_1', A_1'') symmetry. The hypothetical lowest-energy state of $^{14}\text{NH}_3$, with all quanta being zero, has A_1' symmetry. Thus, it is a “missing” state. The *ortho* and *para* nuclear-spin isomers are not linked by any transitions and therefore form distinct (so-called principal) components (PC). The PCs

of the SN can be linked by what we call a “magic number”. In this study, the lowest *ortho* energy level $[(v_1 v_2 v_3^3 v_4^4) J K inv = (0 0 0^0 0^0) 1 1 s]$, see below for the meaning of the label) is fixed to 16,172 993 cm^{-1} , while the energy of the lowest *para* level, $(0 0 0^0 0^0) 0 0 a$, is fixed to 0.793 403 cm^{-1} . Both of these values are taken from Ref. [105]. Note that future improvements of ammonia spectroscopy may result in changes in these values, similar to what happened to the “magic number” of H_2^{16}O [106].

2.1. State labeling

In the study of Al-Derzi et al. [22] each rovibrational eigenstate of $^{14}\text{NH}_3$ was represented by the following set of 12 descriptors: $[v_1 v_2 v_3 v_4 l_3 l_4 J K inv \Gamma_{\text{rot}} \Gamma_{\text{vib}} \Gamma_{\text{tot}}]$, where v_i ($i = 1, \dots, 4$) are the vibrational normal-mode quantum numbers (employing canonical ordering), l_3 and l_4 are the vibrational angular-momentum quantum numbers associated with modes 3 and 4, respectively, J is the total angular-momentum quantum number, $K = |k|$ is the projection of the total angular momentum on the molecule-fixed axis z , and $inv = a/s$ is the inversion symmetry (asymmetric/symmetric or odd/even) of the vibrational motion. In the MARVEL database of Al-Derzi et al. [22], 0 represents the symmetric/even, while 1 the asymmetric/odd choice, while in this study we use s for the symmetric and a for the asymmetric inversion. The last three descriptors of the label provide the rotational (Γ_{rot}), the vibrational (Γ_{vib}), and the full (Γ_{tot}) symmetry of the eigenstate. This information comes from sophisticated quantum-chemical computations of the rovibrational levels and helps when matching empirical energy levels with their first-principles counterparts.

In 2016, several labeling problems of the database of Al-Derzi et al. [22] related to the sign of l_3 and l_4 were identified and corrected by two of the present co-authors [13]. Determination of the sign of l_i is almost impossible, especially in highly-excited cases. Therefore, the L_3 and L_4 values were introduced as quantum numbers used in the label, where $L_i = |l_i|$, as had been recommended previously [107]. During the present study we decided to employ the L_i labels and add an index (or counting number) for levels within each $J - \Gamma_{\text{tot}}$ block of the CoYuTe [71] energy list, N_{block} , to the set of descriptors, since (a) the block index is unique, (b) determination of N_{block} is a preliminary test whether the given empirical energy level exists in the first-principles energy list or not, and (c) this index facilitates the straightforward matching with first-principles energy lists [67,68,108]. At the same time, we remove the redundant Γ_{rot} and Γ_{vib} labels. Thus, in the present MARVEL database of $^{14}\text{NH}_3$ the following 11 descriptors identify a rovibrational state: $[v_1 v_2 v_3 v_4 L_3 L_4 J K inv \Gamma_{\text{tot}} N_{\text{block}}]$.

2.2. Experimental data sources

All sources of experimental rovibrational data collated by Al-Derzi et al. [22], Refs. [16,32,75–81,105,109–154], except 86CoLe [65] (because its wavenumber range begins at 15 259 cm^{-1} , above the range covered in this study), were retained in our compilation. Furthermore, we (re-)evaluated a number of older experimental sources, Refs. [42–64], not used in the final compilation of Al-Derzi et al. [22]. A number of new data sources [30,31,33,34,82] were also included in the revised set of experimental transitions. We ignored six new data sources [35,37–40,155] because either they deal with spectra at wavenumbers above 7500 cm^{-1} [35,39,40], thus they are irrelevant for the present study, or their resolution is not sufficient for our present purpose [37,38,155]. The very recent infrared transmission spectrum of hot ammonia reported by 20BeWoBe [156] was not considered as this source contains no line assignments.

Table 1 contains information about all the data sources included in our present compilation. It is important to reemphasize that

the present MARVEL database contains only those experimentally measured lines in which both energy levels are under 7500 cm^{-1} and are connected to the PCs. As a result, the database assembled contains no floating components (FC). Thus, in this study the experimental SN of $^{14}\text{NH}_3$ contains 15 729 and 30 386 transitions in the *ortho* and *para* PCs, respectively. Table 2 shows the vibrational bands and the number of rovibrational states belonging to each vibrational band origin (VBO) included in the present MARVEL database.

The final MARVEL run of this study considered 46 115 transitions from 82 sources (28 new), of these 46 037 were validated. These lines determine 1753 *ortho* and 3183 *para* energy levels. Files containing our final transition set (wavenumbers of unvalidated lines are indicated with negative sign) and energy levels are deposited in the supplementary data and can also be found on the MARVEL website at <http://kkrk.chem.elte.hu/marvelonline/molecules.php>. By comparison, Al-Derzi et al. [22] validated 27 930 measured transitions characterized by energy values less than 7500 cm^{-1} . Overall, the number of energy levels already present in the analysis of Al-Derzi et al. [22] up to 7500 cm^{-1} , 4716, increases to 4936 at the end of the present study. Thus, although we treated 65% more transitions, the number of energy levels increased only marginally, by 5%. The 220 new energy levels come from only two sources, 146 from Ref. 15BaYuTeCl [32] and 74 from 16PeYuPi [33].

One of the most recent sources of $^{14}\text{NH}_3$ transitions, 16TwHaSe [30], reports 58 highly accurate lines, with uncertainties estimated to be 20–30 kHz, for the $v_1 + v_3$, $v_1 + 2v_4$, and $v_3 + 2v_4$ bands. Now, in the MARVEL world it often happens that different sources with about the same accuracy predict distinctly different empirical energies. However, it is unprecedented that a highly-accurate source is in conflict with other previous measurements and dependable first-principles results at the same time. Two lines of 16TwHaSe show such conflicts with the lines of 12SuBrHuSc [79], 07LiLeXu [145], and 14FoGoHeSo [135]. The first “problematic” line, at 6488.199 520 0 cm^{-1} , predicts an energy value of about 6951.907 cm^{-1} for the $(0 0 1^1 2^2) 66 a$ state, while the previous measurements suggest 6951.940 cm^{-1} . Note that basically the same conflict has already been noted in 12SuBrHuSc [79]. There two out of three relevant lines predicted an energy of 6951.940 cm^{-1} , while the third line yielded 6951.906 cm^{-1} . The second “problematic” line of 16TwHaSe is the one at 6508.300 271 2 cm^{-1} . This line corresponds to an energy of about 6673.631 cm^{-1} for the $(1 0 0^0 2^2) 3 2 s$ state, while the other measurements [135,145] yield 6672.711 cm^{-1} . Since the CoYuTe first-principles energy list seemingly supports the results of the older measurements, we tentatively deleted the two conflicting lines of 16TwHaSe mentioned from the final MARVEL analysis.

2.3. Comparison of empirical energy levels with first-principles data

Al-Derzi et al. [22] partially validated the empirical energy levels of $^{14}\text{NH}_3$ by comparing them with their counterparts in the BYTe variationally computed energy list [67]. The accuracy of variationally computed energies has recently been improved significantly with the production of a ‘spectroscopic’ potential energy surface (PES) called C2018 [70], produced by refining the high-accuracy *ab initio* PES of Polyansky et al. [157] to energies carefully selected from those produced by Al-Derzi et al. [22]. Since the energy list of C2018 contains energies only up to $J = 20$, during this study we utilized the extended version of this database, called CoYuTe [71], which provides energies up to $J = 43$. Nevertheless, one must emphasize that the fitting of the C2018 PES included empirical energies only up to $J = 8$ (to include any higher levels of rotational excitation would have been computationally too expensive).

Table 1

Characteristics of the experimental data sources of rovibrational transitions of $^{14}\text{NH}_3$ analyzed during the present study. A and V stand for the available (below 7500 cm^{-1}) and validated transition entries in a given segment, respectively. ESU, MSU, and LSU designate the estimated, the median, and the largest segment uncertainties, in cm^{-1} , respectively. Bold-face tag entries refer to sources not considered by Al-Derzi et al. [22]. The sources are listed according to their decreasing accuracy in column ESU, with the restriction that segments of the same source are listed consecutively.

Segment tag	Range	A/V	ESU	MSU	LSU
67Kukulich [43]	0.79036–0.79622	3/3	7.00e–09	7.00e–09	7.00e–09
70KuWo [44]	0.72395–0.81833	5/5	7.00e–09	7.00e–09	7.00e–09
09CaDoPu [64]	19.096–19.096	2/2	6.50e–09	6.50e–09	7.00e–09
65Kukulich [42]	0.76167–0.76167	1/1	1.70e–08	1.70e–08	1.70e–08
74CoPo [109]	0.68898–1.9580	50/50	1.67e–07	1.67e–07	1.33e–06
75PoKa [110]	0.24303–1.3323	119/119	1.67e–07	1.67e–07	3.88e–06
80SiSm [50]	0.18123–0.40734	15/15	3.34e–07	3.34e–07	3.34e–07
88TaEnHi [111]	10.670–11.077	14/14	5.31e–07	5.31e–07	5.43e–06
82SaHaAmSh [112]	2.2768–5.9274	12/11	6.67e–07	6.67e–07	6.67e–07
96WiBeKIUr [62]	38.975–40.536	3/3	6.70e–07	6.70e–07	6.70e–07
16TwhaSe [30]	6487.8–6636.7	58/56	6.67e–07	6.67e–07	1.67e–06
98FiKhRule [113]	967.25–967.25	1/1	8.74e–07	8.74e–07	8.74e–07
90SmFiDa [60]	0.33685–1.5385	14/14	1.30e–06	1.30e–06	3.15e–06
80Cohen [114]	2.0862–4.0680	17/17	2.34e–06	2.34e–06	6.67e–06
92SaEnHiPo [75]	2.3527–1387.9	80/80	2.00e–06	2.00e–06	1.89e–05
92SaEnHiPo_S2 [75]	1388.1–1859.7	727/727	2.00e–04	2.00e–04	2.33e–02
10YuPeDrSu [81]	13.334–157.19	175/175	1.70e–06	1.70e–06	6.32e–05
10YuPeDrSu_S2 [81]	39.080–672.64	1600/1600	3.00e–04	3.00e–04	1.73e–02
84MaScFrKr [55]	934.38–1075.2	9/9	3.30e–06	3.30e–06	1.30e–05
08SuLeXu [63]	1027.0–1075.2	14/14	3.34e–06	3.34e–06	1.43e–05
11DrYuPeGu [115]	84.014–89.042	5/5	3.30e–06	3.30e–06	3.30e–06
11LeTrDaBo [16]	965.79–965.79	1/1	3.34e–06	3.34e–06	3.34e–06
06ChPePiMa [80]	15.552–46.468	30/30	3.80e–06	3.80e–06	6.21e–05
96KrTrBoBa [116]	40.523–40.536	2/2	5.30e–06	5.30e–06	5.94e–06
16PeYuPi [33]	13.722–90.955	159/159	6.70e–06	6.70e–06	6.05e–05
16PeYuPi_S2 [33]	39.554–707.26	1221/1221	6.00e–04	6.00e–04	3.81e–02
16PeYuPi_S3 [33]	61.937–682.28	1470/1470	1.00e–03	1.00e–03	3.92e–02
98BeUrWi [117]	4.6746–25.446	3/3	1.11e–05	1.11e–05	2.85e–05
00UrHeKhFi [118]	948.23–951.78	2/2	1.33e–05	1.33e–05	2.64e–05
85SiRe [56]	788.51–1084.6	15/15	1.74e–05	1.74e–05	6.81e–05
80BeGeKrMa [49]	4.6747–35.793	55/55	3.34e–05	3.34e–05	3.13e–04
11GujemoPe [119]	1126.0–1171.4	22/22	4.00e–05	4.00e–05	1.75e–04
82MiToCaLi [52]	1084.6–1084.6	5/5	6.66e–05	6.66e–05	1.09e–04
85BrTo [120]	814.24–1122.1	81/81	7.00e–05	7.01e–05	3.01e–04
86SaScMaPo [121]	772.44–1157.4	138/138	6.79e–05	6.79e–05	1.58e–03
89UrTuRaGu [122]	3984.9–4648.0	785/785	9.00e–05	9.00e–05	9.37e–04
89UrTuRaGu_S2 [122]	3982.8–4633.1	74/74	2.61e–03	2.61e–03	9.58e–03
89UrTuRaGu_S3 [122]	3969.9–4658.6	32/32	1.41e–02	1.41e–02	3.76e–02
77KoMuHiBu [47]	887.88–971.88	11/11	1.41e–04	1.41e–04	3.40e–04
83PoMa [54]	19.096–316.78	204/204	1.28e–04	1.28e–04	4.36e–03
81SaMiWo [123]	932.88–1084.6	32/32	2.00e–04	2.00e–04	8.16e–03
81SaWo [51]	924.95–1043.1	39/39	2.00e–04	2.00e–04	2.00e–04
83ShBjSc [53]	891.88–1103.5	88/88	2.00e–04	2.00e–04	1.17e–03
83UrPaKaYa [124]	709.09–1158.7	640/640	2.00e–04	2.00e–04	2.09e–02
84PoMa [125]	723.27–1250.2	533/533	2.00e–04	2.00e–04	1.03e–02
94ChChCh [126]	928.56–955.06	48/48	2.00e–04	2.00e–04	7.36e–04
95FaltYa [61]	904.80–1178.8	133/133	2.00e–04	2.00e–04	7.88e–03
76FrOk [45]	926.05–1084.6	56/55	2.65e–04	2.65e–04	3.88e–03
79HiJeFa [127]	949.45–949.45	1/1	4.10e–04	4.10e–04	4.10e–04
95KlTaBr [76]	429.37–3114.8	2066/2062	4.00e–04	4.00e–04	3.95e–02
18PeYuPeSu [82]	16.068–1572.8	1266/1266	4.16e–04	4.16e–04	3.07e–02
77HiKoBuFa [46]	887.88–1027.0	9/9	5.00e–04	5.00e–04	6.70e–04
81UrSpPaKa [128]	4.6747–969.02	299/299	4.70e–04	4.70e–04	2.18e–02
84Weber [129]	1511.3–1880.3	176/176	5.00e–04	5.00e–04	8.08e–03
87DCunha [57]	701.46–1187.4	196/196	5.00e–04	5.00e–04	1.72e–02
87LeLaGuTa [58]	1800.0–2099.4	455/455	5.00e–04	5.00e–04	2.49e–02
89GuAbTuRa [130]	3023.3–3674.9	1382/1380	5.00e–04	5.00e–04	2.91e–02
93PiDa [131]	3331.0–3415.4	100/100	5.00e–04	5.00e–04	3.23e–03
96BrMa [132]	4802.2–5293.0	615/610	5.00e–04	5.00e–04	3.15e–02
99FaYa [133]	770.91–1178.8	126/126	5.00e–04	5.00e–04	2.21e–03
99KlBrTaKo [134]	2980.4–3633.8	2162/2162	5.00e–04	5.00e–04	5.21e–02
00CoKlTaBr [77]	1333.0–2096.1	1202/1202	5.00e–04	5.00e–04	2.65e–03
01CoTaKlBr [78]	1201.9–2749.2	1305/1305	5.00e–04	5.00e–04	4.32e–02
13ArMaBo [31]	991.69–1859.7	462/462	5.00e–04	5.00e–04	5.96e–03
14FoGoHeSo [135]	6410.3–6764.5	212/209	5.00e–04	5.00e–04	3.62e–02
86UrCuMaRa [136]	511.37–1347.8	579/579	5.77e–04	5.77e–04	4.15e–02
84UrCuNaPa [105]	1167.9–2126.7	950/949	6.60e–04	6.60e–04	3.74e–02
78Nereson [137]	942.57–956.15	27/27	8.11e–04	8.11e–04	3.95e–03
86PaUrSpRa [138]	1425.8–1892.2	73/70	7.64e–04	7.64e–04	7.71e–03
94BrPe [139]	38.976–3415.1	174/174	9.07e–04	9.07e–04	2.80e–03

(continued on next page)

Table 1 (continued)

Segment tag	Range	A/V	ESU	MSU	LSU
78Jones [48]	887.88–1032.1	23/23	1.00e–03	1.00e–03	3.09e–03
85AnFiFrIl [140]	3134.6–3620.2	615/615	1.37e–03	1.37e–03	3.70e–02
11ZoShOvPo [141]	779.56–2029.2	5915/5900	1.00e–03	1.00e–03	8.19e–02
12SuBrHuSc [79]	6346.8–6973.5	1072/1057	1.00e–03	1.00e–03	4.00e–02
13DoHiYuTe [142]	0.42742–4810.6	10 661/10 657	1.00e–03	1.00e–03	6.27e–02
14CeHoVeCa [143]	4275.4–4339.7	229/228	1.00e–03	1.00e–03	3.89e–02
16SuYuPePi [34]	50.711–657.78	1725/1724	1.00e–03	1.00e–03	3.37e–02
86HeBiBa [144]	620.57–739.44	223/223	1.81e–03	1.81e–03	1.25e–02
07LiLeXu [145]	6421.2–6678.0	97/97	2.00e–03	2.00e–03	3.88e–02
08LeLiXu [146]	6440.0–6832.3	222/221	2.00e–03	2.00e–03	2.13e–02
14DiMiQuSc [147]	3355.0–3355.0	1/1	3.00e–03	3.00e–03	3.00e–03
80UrSpPaMc [148]	678.38–2014.0	482/481	3.91e–03	3.91e–03	5.00e–02
71HeDeGo [149]	19.885–19.885	1/1	5.26e–03	5.26e–03	5.26e–03
85UrMiRa [150]	3148.2–4503.7	478/478	5.00e–03	5.00e–03	3.94e–02
88SnBa [59]	931.18–932.14	7/7	5.00e–03	5.00e–03	5.00e–03
93LuHeNi [151]	6405.5–6887.9	315/314	5.33e–03	5.33e–03	1.70e–02
10PeHa [152]	6460.4–6541.4	16/16	5.00e–03	5.00e–03	1.06e–02
99BePeMe [153]	6540.2–6624.5	27/27	6.14e–03	6.14e–03	1.09e–02
15BaYuTeCl [32]	500.53–1883.7	1366/1349	1.36e–02	1.36e–02	8.08e–02

Fig. 1 shows the differences between the MARVEL and the CoYuTe [71] energies. As seen there, almost all MARVEL energy levels are reproduced by CoYuTe within 0.5 cm^{-1} , proving the high accuracy of the underlying PES. Residuals larger than this are typically associated with the CoYuTe vibrational labels of $2\nu_2 + 3\nu_4$, $4\nu_4$, $\nu_1 + \nu_3$ or $4\nu_2 + 2\nu_4$, which are sometimes in conflict with the labels used in the present MARVEL dataset.

2.4. Comparison of empirical energy levels with existing EH data

Besides the (complete) set of first-principles rovibrational energies, EH energies, characterizing just one or a few vibrational bands, can also be used for validating empirical rovibrational energies derived during a MARVEL analysis. Thus, we employed the EH energy levels of 18PeYuPeSu [82] to check the MARVEL results of this study. As clearly seen in Fig. 2, on the vibrational ground state most of the MARVEL levels can be reproduced by the EH results within 0.005 cm^{-1} . Nevertheless, there are two sets of energy levels where the differences are significantly larger (about 0.015 and 0.15 cm^{-1}). Checking the experimental transitions we could not explain this unusual behavior. Therefore, we decided to compare the

18PeYuPeSu [82] EH energies to the first-principles CoYuTe levels. Fig. 3 presents a subset of the results of this comparison.

As can be seen in Fig. 3, we detect unacceptably large differences at high K values and the systematic, nearly horizontal differences are characterized by always the same K values. Thus, we decided to scrutinize the EH fits of the ground vibrational state and determine the reason behind these large discrepancies, much larger than the proven uncertainty of the first-principles CoYuTe levels (see the previous subsection).

After checking the spectroscopic database of 18PeYuPeSu [82], employed to fit a high-order Watson-type EH model to transitions corresponding to the ground vibrational state, it turned out that each high- K group of outliers, in fact those on the horizontal lines in Fig. 3, forms a separate FC in the experimental SN of $^{14}\text{NH}_3$. This means that although there are measured transitions connecting the energy levels within a FC of the assembled experimental SN of $^{14}\text{NH}_3$, there are no measured transitions which connect to other (high) K values. This in turn means that, at least in the case of $^{14}\text{NH}_3$, including transitions present in FCs in the EH fitting results in unacceptable higher-order EH parameters and energy levels even when all the transitions are fitted well. This means that

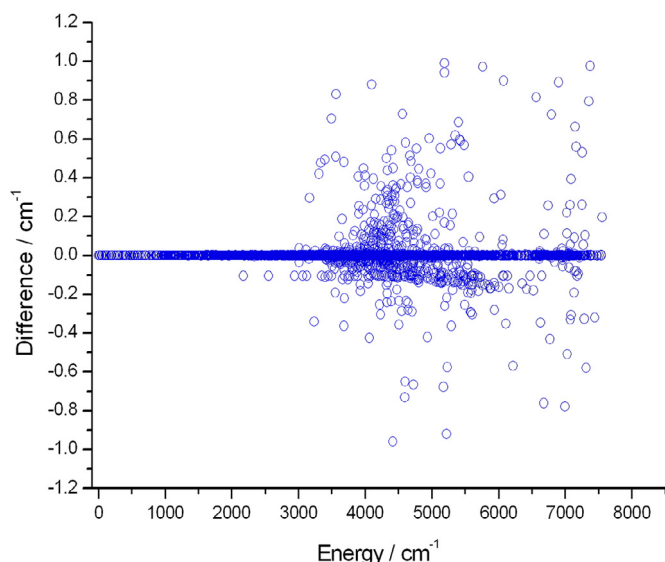


Fig. 1. Comparison of the empirical energies of $^{14}\text{NH}_3$ obtained during this study with their CoYuTe [71] counterparts.

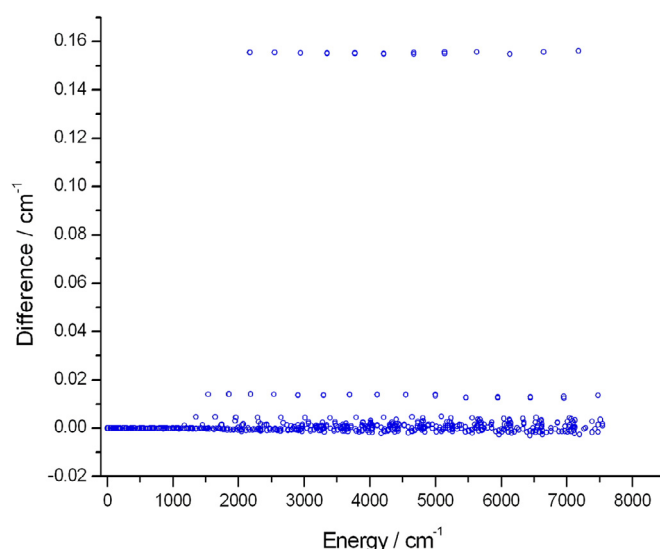


Fig. 2. Comparison of the empirical ground-vibrational-state energies of $^{14}\text{NH}_3$ of this study with their 18PeYuPeSu [82] counterparts.

Table 2

Vibrational band origins (VBO) of $^{14}\text{NH}_3$. VBO values with associated uncertainties corresponding to the last few digits are taken from the present MARVEL database, while those given in brackets correspond to first-principles estimates of Ref. [71]. See text for the meaning of the normal-mode and symmetry labels. Entries in italics correspond to “missing” vibrational energy levels of symmetry $\Gamma_{\text{vib}} = A_1'$. The number of rovibrational levels (RL) related to each vibrational parent state is given in the last column.

ν_1	ν_2	ν_3	ν_4	L_3	L_4	inv	Γ_{vib}	VBO / cm^{-1}	RL
0	0	0	0	0	0	s	A_1'	0.0(0)	351
0	0	0	0	0	0	a	A_2''	0.7934030(10)	353
0	1	0	0	0	0	s	A_1'	[932.4]	305
0	1	0	0	0	0	a	A_2''	968.121930(28)	309
0	2	0	0	0	0	s	A_1'	[1597.5]	193
0	0	0	1	0	1	s	E'	1626.27472(29)	337
0	0	0	1	0	1	a	E''	1627.37238(20)	339
0	2	0	0	0	0	a	A_2''	1882.177445(30)	230
0	3	0	0	0	0	s	A_1'	[2384.2]	107
0	1	0	1	0	1	s	E'	2540.5243(27)	192
0	1	0	1	0	1	a	E''	2586.12855(50)	187
0	3	0	0	0	0	a	A_2''	2895.52192(42)	101
0	2	0	1	0	1	s	E'	[3189.4]	18
0	0	0	2	0	0	s	A_1'	[3215.9]	58
0	0	0	2	0	0	a	A_2''	3217.57919(50)	40
0	0	0	2	0	2	s	E'	3240.16297(53)	116
0	0	0	2	0	2	a	E''	3241.59828(65)	101
1	0	0	0	0	0	s	A_1'	[3336.1]	61
1	0	0	0	0	0	a	A_2''	3337.09694(53)	58
0	0	1	0	1	0	s	E'	3443.62788(54)	124
0	0	1	0	1	0	a	E''	3443.98765(53)	125
0	4	0	0	0	0	s	A_1'	[3462.4]	45
0	2	0	1	0	1	a	E''	[3502.6]	15
0	4	0	0	0	0	a	A_2''	[4061.6]	2
0	1	0	2	0	0	s	A_1'	[4115.9]	4
0	1	0	2	0	2	s	E'	[4135.8]	26
0	1	0	2	0	0	a	A_2''	[4173.1]	12
0	1	0	2	0	2	a	E''	[4192.9]	22
1	1	0	0	0	0	s	A_1'	[4294.5]	68
1	1	0	0	0	0	a	A_2''	4320.030806(90)	68
0	1	1	0	1	0	s	E'	4416.915053(90)	111
0	1	1	0	1	0	a	E''	4435.446476(90)	103
0	2	0	2	0	2	s	E'	[4773.8]	1
0	0	0	3	0	1	s	E'	[4799.2]	2
0	0	0	3	0	1	a	E''	[4801.4]	1
0	0	0	3	0	3	s	A_1'	[4841.6]	1
1	0	0	1	0	1	s	E'	4955.75609(50)	98
1	0	0	1	0	1	a	E''	[4956.9]	54
1	2	0	0	0	0	s	A_1'	[5000.2]	3
0	0	1	1	1	1	s	A_2''	[5052.0]	7
0	0	1	1	1	1	s	E'	[5052.6]	46
0	0	1	1	1	1	a	A_1''	[5052.7]	6
0	0	1	1	1	1	a	E''	[5053.2]	18
0	0	1	1	1	1	a	A_2''	[5067.7]	6
0	0	1	1	1	1	s	A_1'	[5067.8]	9
0	2	0	2	0	0	a	A_2''	[5093.6]	6
0	4	0	1	0	1	s	E'	[5104.9]	2
0	2	1	0	1	0	s	E'	[5144.9]	14
0	2	1	0	1	0	a	E''	[5353.0]	12
0	4	0	1	0	1	a	E''	[5708.3]	1
1	1	0	1	0	1	s	E'	[5897.8]	1
1	0	0	2	0	2	s	E'	6556.4217(12)	39
1	0	0	2	0	2	a	E''	6557.9306(13)	39
1	0	0	2	0	0	s	A_1'	[6603.8]	1
1	0	0	2	0	0	a	A_2''	[6604.7]	1
1	0	1	0	1	0	s	E'	6608.82189(50)	75
1	0	1	0	1	0	a	E''	6609.75336(50)	73
0	0	1	2	1	2	s	A_1'	[6650.8]	8
0	0	1	2	1	2	a	A_1''	[6651.7]	6
0	0	1	2	1	2	a	A_2''	[6652.0]	2
0	0	1	2	1	2	a	E''	6678.31006(50)	64
0	0	1	2	1	2	s	E'	6677.43169(50)	66
0	2	0	3	0	1	a	E''	[6680.4]	1
0	0	2	0	0	0	a	A_2''	6795.3378(16)	29
0	0	2	0	0	0	s	A_1'	[6796.8]	27
0	0	2	0	2	0	s	E'	6850.2449(10)	18
0	0	2	0	2	0	a	E''	6850.6550(10)	18

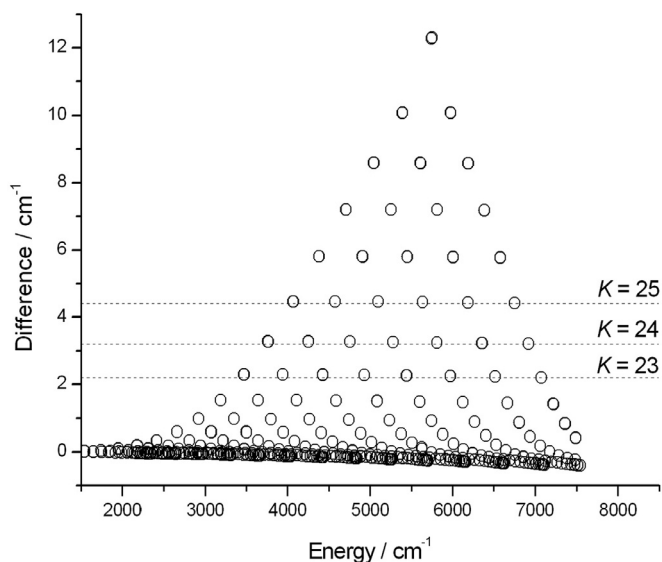


Fig. 3. Effective Hamiltonian model energies of $^{18}\text{PeYuPeSu}$ [82] compared to their CoYuTe [71] first-principles counterparts above $J = 15$ on the ground vibrational state of $^{14}\text{NH}_3$.

high- K -value energy levels ($K > 18$) are ill-determined in the EH fits and as a result the corresponding energy levels derived from the sources $^{10}\text{YuPeDrSu}$ [81] and $^{18}\text{PeYuPeSu}$ [82] are not trustworthy. Thus, we decided to perform a new EH fit on the ground vibrational state using an extended database. The results are described in the next section.

3. New effective Hamiltonian fitting to the pure rotational transitions of $^{14}\text{NH}_3$

As mentioned in the previous section, although earlier EH fits could reproduce all the experimental transitions in the SN of $^{14}\text{NH}_3$ rather well for all vibrational parents tested, the energy values corresponding to states taking part only in FCs appear to be incorrect (see Fig. 3). Our analysis suggested that in order to obtain correct energy values we need to connect the FCs to the PCs (*ortho* and *para*) of the SN. This can be achieved, for example, by adding “artificial” transitions to the SN which connect the FCs to the two PCs. We call these transitions artificial as by design they do not need to obey any selection rules.

To check the effect of the FCs on the empirical energy levels, we created a test dataset containing only measured rotational transitions of the ground vibrational state of $^{14}\text{NH}_3$. This test dataset contains 1660 transitions, 433 of which are distributed among 47 FCs of various size. The smallest J value in the FCs is 13, while the largest one is 35. Most of the FCs consist of states of a given K value with a range of J , and there are only a few cases where the K value is less than $J/2$. The largest FC contains 62 energy levels with the associated transitions having K equal to either 13 or 16 and J in the range from 13 to 30. After employing the $^{10}\text{YuPeDrSu}$ effective Hamiltonian model to these rotational test data, we obtained an EH fit (it is referred as EH Fit M hereafter, with M indicating “measurements only”). The predicted energies from Fit M exhibited basically the same behavior as depicted in Fig. 3. In other words, it became clear that the effective Hamiltonian model cannot reproduce those energy levels of CoYuTe which are part of FCs. As the next step, we decided to connect the FCs to the PCs through first-principles transitions.

By taking advantage of the proven high accuracy of the CoYuTe first-principles energies one can reliably connect the FCs to the appropriate PC. Since the test database contains 47 FCs, we would

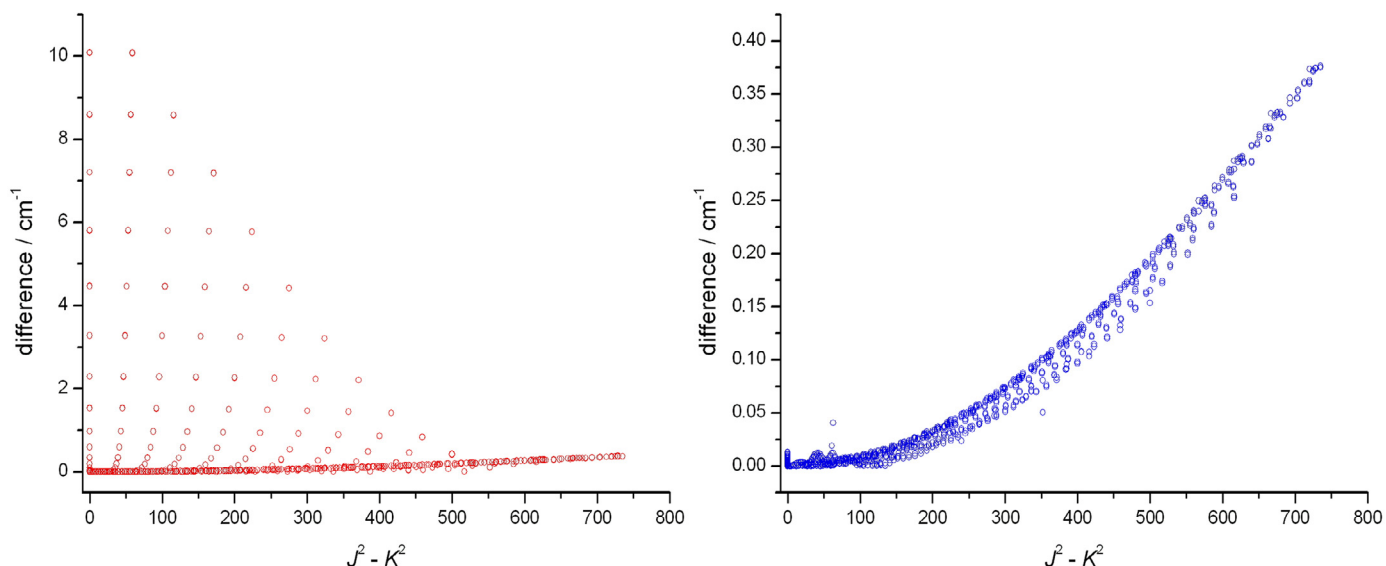


Fig. 4. Effective Hamiltonian model energies of 18PeYuPeSu [82] (left panel) and this work (right panel) compared to their CoYuTe [71] first-principles counterparts on the ground vibrational state of $^{14}\text{NH}_3$ up to 7500 cm^{-1} . Note the very different vertical scale of the two figures.

need at least 47 first-principles “artificial” transitions to eliminate all of the FCs. However, with the minimum number of connections one can not obtain an estimate of the unknown uncertainties of the theoretical lines. Therefore, we added as many computed lines to the test network as was feasible. To determine the uncertainty of a given theoretical transition we calculated the discrepancies of the cycles which contain the given first-principles transition and the absolute average of these discrepancies was accepted as the uncertainty of the first-principles “artificial” transitions. The improved test dataset thus derived for the ground vibrational state of $^{14}\text{NH}_3$ contains 1660 measured and 936 “artificial” lines. We obtained one EH fit to this improved test dataset, referred to as EH Fit A hereafter, with A indicating that “artificial” transitions have been added.

Our new Fit A can reproduce the CoYuTe energy levels almost perfectly. The left panel of Fig. 4 shows the differences between the 18PeYuPeSu [82] and the CoYuTe [71] energies, while the right panel shows the same differences using the result of Fit A based on the improved test dataset. As can be seen, each CoYuTe energy level can be reproduced by the new effective Hamiltonian model within 0.4 cm^{-1} all the way to 7500 cm^{-1} .

It is important to note that the Fit A model shows systematic deviations from the CoYuTe levels above $J = 30$. Nevertheless, since the CoYuTe energy levels were generated using the C2018 PES [41], for which the *ab initio* PES was adjusted using the experimental results only up to $J = 8$, it is hard to decide which theoretical model, the first-principles or the effective one, results in “incorrect” energy levels. Further experimental and modeling studies are needed to settle this issue.

Table 3 lists the parameters obtained from our new EH Fit A, based on all available experimental transitions as well as 936 CoYuTe first-principles energy values, our new EH Fit M, based on all available experimental transitions, and the EH fit in 10YuPeDrSu [81], based on all available experimental data in the GS (s), GS (a), ν_2 (s), and ν_2 (a), where GS refers to the ground vibrational state. The fitting input and output files are provided in the Supplementary Material, including the energy levels predicted from the new EH models. Note that Fit M was started with constants from 10YuPeDrSu [81]. Then Fit A was executed with the Fit M final constants as the initial constants, and with the CoYuTe energies with $J \geq 19$ of all K added to the fit to update the EH pa-

rameters. The Fit A parameters are updated by adding the CoYuTe energies one K at a time, and the maximum J and K values of the first-principles energies are 35. The fitting process was straightforward, and we did not experience convergence issues. Note that the 18PeYuPeSu EH model deals with higher vibrational states and its ground states’ constants are fixed at those values determined from 10YuPeDrSu. Thus, the 18PeYuPeSu and 10YuPeDrSu models are exactly the same for the ground state. Here we used the 10YuPeDrSu constants as the initial constants for Fit M, because the uncertainties of the ground state constants are provided in 10YuPeDrSu but not in 18PeYuPeSu.

As shown in Table 3, the same set of 10YuPeDrSu parameters [81] is sufficient to fit the added high J and K CoYuTe energies, and additional higher-order EH parameters are not needed. When comparing EH Fit M to 10YuPeDrSu [81] (see Diff. in Column 6 of Table 3), less than 3% changes are seen in the rotational constants as well as the fourth- (D) and sixth-order (H) centrifugal-distortion constants. Up to 72% changes are seen in the eighth- and higher-order centrifugal-distortion parameters, with the exception of a 130% change in the sixteenth-order centrifugal-distortion constant P_J . Such large changes are not unexpected. NH_3 is a non-rigid molecule, and its EH model has multiple local minima when fitting to a limited set of experimental transitions that can not well constrain the model. When the experimental dataset is changed, the fitting can find another local minimum and significantly change the values of some of the EH parameters. For the levels that are experimentally observed and connected to a PC, the predicted energies from various sets of EH parameter values are expected to agree with each other; for the levels that are not experimentally observed or connected to a PC, the predicted energies may vary significantly among various sets of EH parameters. The CoYuTe first-principles energies utilized connect levels with K and J up to 35 to the PCs and constrain the EH Fit A model with the maximum number of transitions. Thus, the energies predicted from our new EH model with K and J up to 35 are not extrapolated ones.

Direct comparison of EH Fit A and EH Fit M provides useful insight into which terms might be significant and lead to discrepancies at high K values, as the difference between the two fits is only the presence of 936 CoYuTe first-principles high- K energies in Fit A. The CoYuTe first-principles energies utilized connect levels with K and J up to 35 to the PCs and constrain the Fit A model

Table 3

Spectroscopic parameters, in MHz, for the vibrational ground state of $^{14}\text{NH}_3$, where OC is Operator Code in SPFIT and SPCAT, with the last two digits denoting vibrational identifier V1V2, 00 for GS (s), 11 for GS (a), and 01 for interaction between GS (s) and GS (a) (see Ref. [81] for details). There is no well-established naming convention for the higher-order (above octic) spectroscopic constants, the one used here is the one we prefer. Diff. in Columns 4 and 6 correspond to (FitA – FitM)/FitM (FitM – 10YuPeDrSu)/10YuPeDrSu, respectively.

OC	Parameter	New fit A	Diff. (%)	New fit M	Diff. (%)	10YuPeDrSu [81]
0	E	-23785.8904(33)	0.0	-23785.8919(33)	0.0	-23785.895135(112)
100	B	298192.9106(32)	0.0	298192.9067(32)	0.0	298192.92072(104)
1000	$C - B$	-111497.0714(73)	0.0	-111497.0727(74)	0.0	-111497.0613(50)
200	$-D_J$	-25.461858(154)	0.0	-25.461567(160)	0.0	-25.461658(115)
1100	$-D_{JK}$	47.27987(56)	0.0	47.28043(58)	0.0	47.27790(62)
2000	$-D_K$	-27.35564(51)	0.0	-27.35587(59)	0.0	-27.35488(67)
300	H_J	$7.2210(74) \times 10^{-3}$	0.0	$7.2226(79) \times 10^{-3}$	0.5	$7.1836(90) \times 10^{-3}$
1200	H_{JK}	-0.022238(62)	-0.4	-0.022331(65)	1.5	-0.021997(80)
2100	H_{JKK}	0.023737(102)	-0.8	0.023940(107)	2.0	0.023479(133)
3000	H_K	$-8.233(48) \times 10^{-3}$	-1.9	$-8.389(53) \times 10^{-3}$	2.9	$-8.154(63) \times 10^{-3}$
400	L_J	$-3.7693(152) \times 10^{-6}$	0.5	$-3.7503(166) \times 10^{-6}$	1.0	$-3.7140(141) \times 10^{-6}$
1300	L_{JKK}	$1.88(34) \times 10^{-6}$	-45.3	$3.44(37) \times 10^{-6}$	-35.5	$5.333(279) \times 10^{-6}$
2200	L_{JKKK}	$0.04345(168) \times 10^{-3}$	19.4	$0.03639(182) \times 10^{-3}$	40.3	$0.02594(136) \times 10^{-3}$
3100	L_{JKKKK}	$-0.07820(238) \times 10^{-3}$	13.7	$-0.06875(256) \times 10^{-3}$	27.4	$-0.05395(194) \times 10^{-3}$
4000	LK	$0.03683(104) \times 10^{-3}$	9.0	$0.03378(120) \times 10^{-3}$	30.0	$0.02599(86) \times 10^{-3}$
500	M_J	$2.803(38) \times 10^{-9}$	3.5	$2.709(40) \times 10^{-9}$	1.4	$2.671(33) \times 10^{-9}$
1400	M_{JKK}	$0.01556(61) \times 10^{-6}$	21.7	$0.01279(68) \times 10^{-6}$	10.7	$0.01155(49) \times 10^{-6}$
2300	M_{JKKK}	$-0.1885(42) \times 10^{-6}$	7.8	$-0.1748(45) \times 10^{-6}$	3.2	$-0.1693(36) \times 10^{-6}$
3200	M_{JKKKK}	$0.4562(100) \times 10^{-6}$	4.5	$0.4367(103) \times 10^{-6}$	1.0	$0.4325(87) \times 10^{-6}$
4100	M_{JKKKKK}	$-0.4214(99) \times 10^{-6}$	1.9	$-0.4136(100) \times 10^{-6}$	-0.7	$-0.4167(85) \times 10^{-6}$
5000	M_K	$0.1347(34) \times 10^{-6}$	5.1	$0.1282(41) \times 10^{-6}$	-11.5	$0.14483(292) \times 10^{-6}$
600	N_J	$-1.619(57) \times 10^{-12}$	8.7	$-1.489(59) \times 10^{-12}$	6.4	$-1.400(51) \times 10^{-12}$
1500	N_{JKK}	$-4.21(72) \times 10^{-12}$	63.2	$-2.58(80) \times 10^{-12}$	-9.2	$-2.84(51) \times 10^{-12}$
2400	N_{JKKK}	$0.2249(72) \times 10^{-9}$	11.1	$0.2025(77) \times 10^{-9}$	2.2	$0.1981(57) \times 10^{-9}$
3300	N_{JKKKK}	$-0.8932(260) \times 10^{-9}$	6.5	$-0.8390(269) \times 10^{-9}$	-1.9	$-0.8555(225) \times 10^{-9}$
4200	N_{JKKKKK}	$1.285(41) \times 10^{-9}$	0.9	$1.273(42) \times 10^{-9}$	-8.4	$1.390(35) \times 10^{-9}$
5100	$N_{JKKKKKK}$	$-0.759(33) \times 10^{-9}$	-6.8	$-0.814(34) \times 10^{-9}$	-17.0	$-0.9811(241) \times 10^{-9}$
6000	N_K	$0.1480(106) \times 10^{-9}$	-24.6	$0.1964(92) \times 10^{-9}$	-11.3	$0.2214(60) \times 10^{-9}$
700	O_J	$0.567(43) \times 10^{-15}$	19.1	$0.476(44) \times 10^{-15}$	21.4	$0.392(39) \times 10^{-15}$
1600	O_{JKK}	$-7.04(52) \times 10^{-15}$	-3.8	$-7.32(55) \times 10^{-15}$	-3.7	$-7.60(45) \times 10^{-15}$
2500	O_{JKKK}	$-0.1518(72) \times 10^{-12}$	15.4	$-0.1315(76) \times 10^{-12}$	-1.5	$-0.1335(53) \times 10^{-12}$
3400	O_{JKKKK}	$0.904(41) \times 10^{-12}$	4.6	$0.864(40) \times 10^{-12}$	-12.9	$0.9923(257) \times 10^{-12}$
4300	O_{JKKKKK}	$-1.716(97) \times 10^{-12}$	-4.9	$-1.805(97) \times 10^{-12}$	-22.9	$-2.341(51) \times 10^{-12}$
5200	$O_{JKKKKKK}$	$1.398(109) \times 10^{-12}$	-16.1	$1.666(124) \times 10^{-12}$	-32.9	$2.482(48) \times 10^{-12}$
6100	$O_{JKKKKKKK}$	$-0.468(57) \times 10^{-12}$	-31.0	$-0.678(76) \times 10^{-12}$	-43.6	$-1.2015(217) \times 10^{-12}$
7000	O_K	$0.0404(105) \times 10^{-12}$	-50.4	$0.0815(232) \times 10^{-12}$	-68.2	$0.2561(46) \times 10^{-12}$
800	P_J	$-0.0773(125) \times 10^{-18}$	51.0	$-0.0512(129) \times 10^{-18}$	130.6	$-0.0222(115) \times 10^{-18}$
1700	P_{JKK}	$2.431(165) \times 10^{-18}$	-1.2	$2.461(174) \times 10^{-18}$	-2.8	$2.531(162) \times 10^{-18}$
2600	P_{JKKK}	$0.0519(33) \times 10^{-15}$	13.6	$0.0457(33) \times 10^{-15}$	-15.7	$0.05422(197) \times 10^{-15}$
3500	P_{JKKKK}	$-0.3612(225) \times 10^{-15}$	1.3	$-0.3564(226) \times 10^{-15}$	-25.0	$-0.4754(106) \times 10^{-15}$
4400	P_{JKKKKK}	$0.795(61) \times 10^{-15}$	-8.5	$0.869(67) \times 10^{-15}$	-34.0	$1.3174(244) \times 10^{-15}$
5300	$P_{JKKKKKK}$	$-0.768(76) \times 10^{-15}$	-18.8	$-0.946(99) \times 10^{-15}$	-43.6	$-1.6766(280) \times 10^{-15}$
6200	$P_{JKKKKKKK}$	$0.322(45) \times 10^{-15}$	-31.0	$0.467(75) \times 10^{-15}$	-55.1	$1.0391(163) \times 10^{-15}$
7100	$P_{JKKKKKKKK}$	$-0.0424(104) \times 10^{-15}$	-46.0	$-0.0785(260) \times 10^{-15}$	-72.3	$-0.2832(43) \times 10^{-15}$
60000	η	$-0.1631(33) \times 10^{-3}$	2.3	$-0.1595(35) \times 10^{-3}$	-11.1	$-0.1794(42) \times 10^{-3}$
60100	η_J	$0.17661(298) \times 10^{-6}$	4.8	$0.1685(36) \times 10^{-6}$	-8.9	$0.1850(38) \times 10^{-6}$
61000	η_K	$-0.3805(172) \times 10^{-6}$	2.1	$-0.3727(196) \times 10^{-6}$	-13.3	$-0.430(32) \times 10^{-6}$
111	V	298041.3739(32)	0.0	298041.3695(32)	0.0	298041.38282(104)
1011	$C - B$	-111285.5847(73)	0.0	-111285.5855(74)	0.0	-111285.5732(50)
211	$-D_J$	-24.956678(153)	0.0	-24.956360(158)	0.0	-24.956414(113)
1111	$-D_{JK}$	45.88634(55)	0.0	45.88684(57)	0.0	45.88426(61)
2011	$-D_K$	-26.40108(50)	0.0	-26.40127(58)	0.0	-26.40027(66)
311	H_J	$6.0455(71) \times 10^{-3}$	0.0	$6.0463(75) \times 10^{-3}$	0.6	$6.0083(86) \times 10^{-3}$
1211	H_{JK}	-0.017421(61)	-0.5	-0.017511(64)	1.9	-0.017179(78)
2111	H_{JKK}	0.017211(101)	-1.1	0.017409(106)	2.7	0.016947(131)
3011	H_K	$-5.307(47) \times 10^{-3}$	-2.8	$-5.460(52) \times 10^{-3}$	4.5	$-5.224(63) \times 10^{-3}$
411	L_J	$-1.6434(144) \times 10^{-6}$	1.6	$-1.6172(155) \times 10^{-6}$	2.1	$-1.5843(127) \times 10^{-6}$
1311	L_{JK}	$-9.48(33) \times 10^{-6}$	18.8	$-7.98(36) \times 10^{-6}$	30.6	$-6.112(270) \times 10^{-6}$
2211	L_{JKK}	$0.06616(166) \times 10^{-3}$	11.7	$0.05922(180) \times 10^{-3}$	21.1	$0.04890(134) \times 10^{-3}$
3111	L_{JKKK}	$-0.09829(236) \times 10^{-3}$	10.5	$-0.08895(254) \times 10^{-3}$	19.7	$-0.07428(192) \times 10^{-3}$
4011	LK	$0.04347(103) \times 10^{-3}$	7.5	$0.04044(119) \times 10^{-3}$	23.7	$0.03269(85) \times 10^{-3}$
511	M_J	$-0.241(38) \times 10^{-9}$	-33.8	$-0.364(39) \times 10^{-9}$	-9.5	$-0.402(33) \times 10^{-9}$
1411	M_{JKK}	$0.03489(58) \times 10^{-6}$	7.7	$0.03239(64) \times 10^{-6}$	3.3	$0.03135(45) \times 10^{-6}$
2311	M_{JKKK}	$-0.2381(41) \times 10^{-6}$	5.7	$-0.2252(43) \times 10^{-6}$	2.1	$-0.2205(35) \times 10^{-6}$
3211	M_{JKKKK}	$0.5204(98) \times 10^{-6}$	3.7	$0.5020(101) \times 10^{-6}$	0.6	$0.4988(85) \times 10^{-6}$
4111	M_{JKKKKK}	$-0.4632(97) \times 10^{-6}$	1.6	$-0.4560(98) \times 10^{-6}$	-0.8	$-0.4596(83) \times 10^{-6}$
5011	M_K	$0.1456(34) \times 10^{-6}$	4.5	$0.1393(40) \times 10^{-6}$	-10.7	$0.15593(286) \times 10^{-6}$
611	N_J	$1.583(57) \times 10^{-12}$	-10.6	$1.771(59) \times 10^{-12}$	-5.4	$1.872(52) \times 10^{-12}$
1511	N_{JKK}	$-0.02662(69) \times 10^{-9}$	3.7	$-0.02566(76) \times 10^{-9}$	-2.5	$-0.02633(49) \times 10^{-9}$
2411	N_{JKKK}	$0.2919(69) \times 10^{-9}$	7.4	$0.2719(73) \times 10^{-9}$	1.0	$0.2691(53) \times 10^{-9}$

(continued on next page)

Table 3 (continued)

OC	Parameter	New fit A	Diff. (%)	New fit M	Diff. (%)	10YuPeDrSu [81]
3311	N_{JJJKK}	$-1.0030(253) \times 10^{-9}$	5.3	$-0.9527(261) \times 10^{-9}$	-1.9	$-0.9712(214) \times 10^{-9}$
4211	N_{JJKKKK}	$1.389(41) \times 10^{-9}$	0.7	$1.380(41) \times 10^{-9}$	-7.9	$1.498(34) \times 10^{-9}$
5111	N_{JKKKKK}	$-0.812(33) \times 10^{-9}$	-6.5	$-0.868(34) \times 10^{-9}$	-16.2	$-1.0362(234) \times 10^{-9}$
6011	N_K	$0.1596(106) \times 10^{-9}$	-23.2	$0.2078(91) \times 10^{-9}$	-10.9	$0.2332(59) \times 10^{-9}$
711	O_J	$-1.527(43) \times 10^{-15}$	-8.9	$-1.676(44) \times 10^{-15}$	-5.5	$-1.773(39) \times 10^{-15}$
1611	O_{JJJK}	$8.36(51) \times 10^{-15}$	-5.1	$8.81(56) \times 10^{-15}$	-0.1	$8.82(48) \times 10^{-15}$
2511	O_{JJJKK}	$-0.2018(70) \times 10^{-12}$	9.4	$-0.1844(73) \times 10^{-12}$	-1.5	$-0.1872(49) \times 10^{-12}$
3411	O_{JJJKKK}	$1.001(41) \times 10^{-12}$	3.7	$0.965(39) \times 10^{-12}$	-11.7	$1.0928(242) \times 10^{-12}$
4311	O_{JJKKKK}	$-1.843(97) \times 10^{-12}$	-4.7	$-1.933(97) \times 10^{-12}$	-21.5	$-2.463(49) \times 10^{-12}$
5211	O_{JKKKKK}	$1.510(110) \times 10^{-12}$	-14.8	$1.772(124) \times 10^{-12}$	-31.3	$2.580(46) \times 10^{-12}$
6111	$O_{JKKKKKK}$	$-0.528(57) \times 10^{-12}$	-27.9	$-0.732(76) \times 10^{-12}$	-41.3	$-1.2480(211) \times 10^{-12}$
7011	O_K	$0.0547(106) \times 10^{-12}$	-41.8	$0.0940(230) \times 10^{-12}$	-64.7	$0.2663(45) \times 10^{-12}$
811	P_J	$0.5354(127) \times 10^{-18}$	-8.2	$0.5830(132) \times 10^{-18}$	-5.2	$0.6153(119) \times 10^{-18}$
1711	P_{JJJK}	$-2.146(173) \times 10^{-18}$	-10.8	$-2.407(187) \times 10^{-18}$	0.9	$-2.385(178) \times 10^{-18}$
2611	P_{JJJKK}	$0.0671(33) \times 10^{-15}$	8.2	$0.0620(33) \times 10^{-15}$	-12.0	$0.07042(181) \times 10^{-15}$
3511	P_{JJJKKK}	$-0.3941(228) \times 10^{-15}$	0.9	$-0.3905(228) \times 10^{-15}$	-23.0	$-0.5071(99) \times 10^{-15}$
4411	$P_{JJJKKKK}$	$0.849(61) \times 10^{-15}$	-7.8	$0.921(67) \times 10^{-15}$	-32.4	$1.3620(233) \times 10^{-15}$
5311	$P_{JJKKKKK}$	$-0.832(77) \times 10^{-15}$	-17.0	$-1.003(99) \times 10^{-15}$	-41.8	$-1.7225(271) \times 10^{-15}$
6211	$P_{JJKKKKKK}$	$0.366(46) \times 10^{-15}$	-27.5	$0.505(75) \times 10^{-15}$	-52.7	$1.0683(160) \times 10^{-15}$
7111	$P_{JKKKKKKK}$	$-0.0552(105) \times 10^{-15}$	-38.5	$-0.0897(258) \times 10^{-15}$	-69.2	$-0.2914(42) \times 10^{-15}$
60,011	η	$-0.1628(34) \times 10^{-3}$	2.3	$-0.1592(36) \times 10^{-3}$	-11.4	$-0.1797(43) \times 10^{-3}$
60,111	η_J	$0.17669(305) \times 10^{-6}$	4.9	$0.1685(37) \times 10^{-6}$	-9.3	$0.1857(39) \times 10^{-6}$
61,011	η_K	$-0.4198(163) \times 10^{-6}$	1.6	$-0.4133(187) \times 10^{-6}$	-8.8	$-0.4533(315) \times 10^{-6}$
230001	α	$-8.650(45)$	0.4	$-8.614(48)$	-3.2	$-8.896(54)$
230101	α_J	$5.6508(299) \times 10^{-3}$	1.5	$5.566(40) \times 10^{-3}$	-2.6	$5.716(38) \times 10^{-3}$
230201	α_{JJ}	$-1.7587(73) \times 10^{-6}$	1.8	$-1.7276(119) \times 10^{-6}$	2.9	$-1.6791(112) \times 10^{-6}$
230301	α_{JJJ}	$1.0305(132) \times 10^{-9}$	7.7	$0.9566(209) \times 10^{-9}$	0.1	$0.9558(157) \times 10^{-9}$
231001	α_K	$-0.03343(75)$	9.0	$-0.03066(81)$	20.9	$-0.02536(54)$
231101	α_{JK}	$0.06667(94) \times 10^{-3}$	6.8	$0.06245(107) \times 10^{-3}$	3.4	$0.06039(76) \times 10^{-3}$
231201	α_{JKK}	$-0.02862(52) \times 10^{-6}$	4.8	$-0.02731(62) \times 10^{-6}$	-6.2	$-0.029115(303) \times 10^{-6}$
232101	α_{JKK}	$0.03176(131) \times 10^{-6}$	-12.0	$0.03608(168) \times 10^{-6}$	-27.6	$0.04986(32) \times 10^{-6}$
232001	α_{KK}	$-0.03217(149) \times 10^{-3}$	-18.3	$-0.03936(177) \times 10^{-3}$	-25.1	$-0.05257(45) \times 10^{-3}$
233001	α_{KKK}	$-1.51(44) \times 10^{-9}$	-37.1	$-2.40(97) \times 10^{-9}$	-69.5	$-7.858(89) \times 10^{-9}$

with a large number of transitions. Thus, we expect improved uncertainties for the terms leading to the discrepancies at high K . In fact, the parameters O_K and $P_{JKKKKKK}$ have the most reduced uncertainties, with uncertainty values in EH Fit A being $\sim 40\%$ and $\sim 45\%$ of those in Fit M. In addition, the following terms are determined with significantly reduced uncertainties: $P_{JJJKKKK}$ ($\sim 60\%$), $P_{JJJKKKK}$ ($\sim 75\%$), and $O_{JJJKKKK}$ ($\sim 75\%$). The rest of the centrifugal-distortion constants changed by less than 10%. Our conclusion is that the constants mentioned play the most significant role in the discrepancies observed at high K values.

A referee of this paper called our attention to a study of Costain [74]. Costain fitted microwave ammonia transitions using the empirical formula

$$\nu = \nu_0 \exp[AJ(J+1) + BK^2 + Cj^2(J+1)^2 + Dj(J+1)K^2 + EK^4] \quad (1)$$

and suggested that the inversion spectrum might be more accurately represented by this formula than by the usual expression used in this work and 10YuPeDrSu [81]. Table S1 of the Supplementary Material lists the lines fitted by Costain [74], the fitting residuals obtained by Costain [74] and those characterizing our EH models. The comparison of fitting residuals indicates that the lines available to Costain are represented better by our EH models. This may be due to the fact that we include many more high J and K lines in our EH models than the 64 lines Costain [74] dealt with. Nevertheless, in a future work we plan to incorporate this empirical formula to fit the data included in EH Fit A and establish the performance of this model.

4. Highly accurate rotational lines of $^{14}\text{NH}_3$

A common feature of most present-day line-by-line spectroscopic databases, including HITRAN 2016 [158], is that they are based almost exclusively on lines obtained from Doppler-broadened spectra. The usual Doppler linewidth is about 1–

5 GHz, strongly limiting the accuracy of the high-resolution lines measured and the resolution of the experiments. Modern laser-spectroscopy measurements, based on frequency combs, cavity enhancement, saturation, and other recent experimental advances, allow Doppler-free measurement of certain lines. The accuracy of these measurements approach the 1 kHz region [106], though in principle it can be even better.

Several scientific and engineering applications need the protection of certain narrow regions around lines so that observations of these molecular transitions, perhaps of astronomical interest, will not be hindered by man-made activities. Due to an effort of a number of international organizations over several decades, a number of lines with frequencies less than 3 THz for a number of molecules, including ammonia, have become protected [159]. The $^{14}\text{NH}_3$ lines under protection by certain international bodies associated with the National Academies of Sciences, Engineering, and Medicine (NASEM) [159] are listed in Table 4.

As shown in Table 4, our MARVEL database contains several sources reporting measurements of the lines protected with an accuracy down to 0.2 kHz. The present MARVEL analysis of all available high-resolution spectroscopic information for $^{14}\text{NH}_3$ confirms the accuracy of the best measurements. Furthermore, in a few cases MARVEL even improves upon the accuracy of the best measurements. In fact, above 2 THz there are several lines, given in Table 4 as boldface entries under $f(\text{MARVEL})$, where the empirical estimates of the present study have tighter uncertainties than the best measurements. Fig. 5 illustrates two such cases. Finally, we note that the designation of quite a few lines above 1 THz used by NASEM [159] are incorrect by more than 2 MHz. These entries are italicized in the first column of Table 4. There are three cases, the lines at 2405.121, 2991.555, and 2994.786 GHz, where both the NASEM designations are incorrect and we obtain higher accuracy as a result of our MARVEL analysis than the best measure-

Table 4

Rotational lines of $^{14}\text{NH}_3$ under protection by international bodies associated with the National Academies of Sciences, Engineering, and Medicine (NASEM) [159] and their various experimental and empirical determinations. All values are in GHz and all lines belong to the vibrational ground state. For each line all measurements treated during the MARVEL analysis are reported. A NASEM entry in italics indicates that the difference between MARVEL and NASEM is larger than 2 MHz. Boldface MARVEL entries correspond to empirical predictions more accurate than the most precise direct measurements.

$f(\text{NASEM})/\text{GHz}$	$J' K' inv' \leftarrow J'' K'' inv''$	$f(\text{MARVEL})/\text{GHz}$	$f(\text{expt})/\text{GHz}$
23.694	1 1 a \leftarrow 1 1 s	23.694 495 49(21)	23.694 495 49(21) [43] 23.694 496 0(50) [110]
23.723	2 2 a \leftarrow 2 2 s	23.722 633 32(21)	23.722 633 32(21) [43] 23.722 631 0(50) [110]
23.870	3 3 a \leftarrow 3 3 s	23.870 129 17(21)	23.870 129 17(21) [43] 23.870 130 0(50) [110]
572.498	1 0 s \leftarrow 0 0 a	572.498 159 87(24)	572.498 159 93(18) [64] 572.498 159 78(21) [64] 572.498 1(24) [54] 572.498 1(30) [128]
1168.452	2 1 s \leftarrow 1 1 a	1168.452 397(23)	1168.452 394(20) [62] 1168.452 413(51) [81] 1168.451 6(24) [54] 1168.477(26) [128] 1168.480(29) [139]
1214.859	2 0 a \leftarrow 1 0 s	1214.852 942(20)	1214.852 942(20) [62] 1214.852 87(14) [116] 1214.858 6(59) [54] 1214.846(18) [139]
1215.245	2 1 a \leftarrow 1 1 s	1215.245 711(23)	1215.245 714(20) [62] 1215.245 67(18) [116] 1215.245 3(24) [54] 1215.272(27) [139]
1763.525	3 0 s \leftarrow 2 0 a	1763.524 359(99)	1763.524 358(99) [81] 1763.525 2(24) [54] 1763.544(21) [139]
1763.602	3 1 s \leftarrow 2 1 a	1763.601 227(81)	1763.601 219(99) [81] 1763.602 0(24) [54]
1763.821	3 2 s \leftarrow 2 2 a	1763.823 18(13)	1763.823 186(99) [81] 1763.821 4(24) [54] 1763.778(47) [139]
1808.936	3 1 a \leftarrow 2 1 s	1808.934 551(81)	1808.934 564(99) [81] 1808.935 4(24) [54] 1808.936(18) [139] 1808.969(36) [128]
1810.378	3 2 a \leftarrow 2 2 s	1810.380 00(13)	1810.379 971(99) [81] 1810.377 7(24) [54] 1810.369(18) [139] 1810.357(24) [128]
2357.210	4 1 s \leftarrow 3 1 a	2357.209 99(10)	2357.209 920(99) [81] 2357.210 2(24) [54] 2357.199(18) [139]
2357.727	4 2 s \leftarrow 3 2 a	2357.728 46(13)	2357.728 417(99) [81] 2357.726 8(24) [54] 2357.697(33) [139]
2358.563	4 3 s \leftarrow 3 3 a	2358.566 19(17)	2358.566 118(99) [81] 2358.563 2(31) [54] 2358.551(18) [139]
2400.018	4 0 a \leftarrow 3 0 s	2400.020 831(99)	2400.020 831(99) [81] 2400.017 6(34) [54] 2400.024(18) [139]
2400.578	4 1 a \leftarrow 3 1 s	2400.578 800(78)	2400.578 775(99) [81] 2400.578 5(24) [54] 2400.573(18) [139]
2402.265	4 2 a \leftarrow 3 2 s	2402.266 00(13)	2402.266 037(99) [81] 2402.264 8(24) [54] 2402.252(18) [139]
2405.121	4 3 a \leftarrow 3 3 s	2405.124 63(17)	2405.121 3(35) [54] 2405.118(18) [139]
2948.411	5 0 s \leftarrow 4 0 a	2948.395 32(14)	2948.395 304(99) [81] 2948.411(16) [54] 2948.375(21) [139]
2948.669	5 1 s \leftarrow 4 1 a	2948.668 64(13)	2948.668 655(99) [81] 2948.669 3(24) [54] 2948.711(44) [139]
2949.480	5 2 s \leftarrow 4 2 a	2949.482 41(16)	2949.482 400(99) [81] 2949.480 5(24) [54] 2949.457(27) [139]

(continued on next page)

Table 4 (continued)

$f(\text{NASEM})/\text{GHz}$	$J' K' inv' \leftarrow J'' K'' inv''$	$f(\text{MARVEL})/\text{GHz}$	$f(\text{expt})/\text{GHz}$
2950.815	5 3 s \leftarrow 4 3 a	2950.819 80(13)	2950.819 774(99) [81] 2950.814 6(54) [54] 2950.800(21) [139]
2952.640	5 4 s \leftarrow 4 4 a	2952.638 37(10)	2952.638 366(99) [81] 2952.639 7(24) [54] 2952.632(18) [139]
2989.643	5 1 a \leftarrow 4 1 s	2989.641 29(12)	2989.642 5(24) [54] 2989.641 3(30) [128] 2989.632(18) [139]
2991.555	5 2 a \leftarrow 4 2 s	2991.557 32(16)	2991.555 2(24) [54] 2991.533(26) [139] 2991.527(32) [128]
2994.786	5 3 a \leftarrow 4 3 s	2994.793 39(13)	2994.786 4(73) [54] 2994.780(18) [139] 2994.822(30) [128]
2999.430	5 4 a \leftarrow 4 4 s	2999.430 81(10)	2999.430 1(24) [54] 2999.418(14) [128] 2999.400(33) [139]

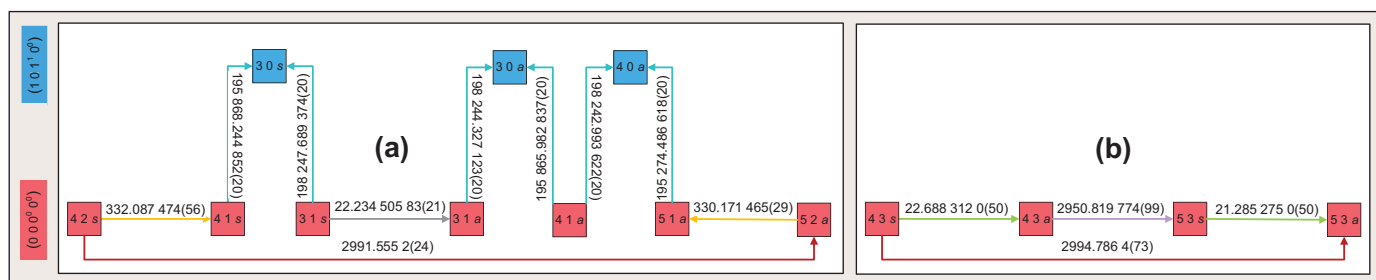


Fig. 5. Highly accurate frequency determinations for lines at 2991.555 [panel (a)] and 2994.786 GHz [panel (b)] NASEM frequencies (see also Table 4). The $J K$ rotational labels within the squares represent rovibrational states, whose $(v_1 v_2 v_3 v_4)$ vibrational labels are indicated in the left-hand-side legend with different colors. Transitions designated with orange, cyan, gray, brown, green, and purple arrows are results from 88TaEnHi [111], 16TwHaSe [30], 70KuWo [44], 83PoMa [54], 75PoKa [110], and 10YuPeDrSu [81], respectively. Lines are augmented with their experimental frequencies (in GHz) and the uncertainties of the last few frequency digits given in parentheses. In both cases (a) and (b), the brown connectors of 83PoMa denote the best available observations for the corresponding transitions. Employing the Ritz principle successively, the following empirical predictions are derived from the two underlying network paths: $332.087\,474 + 195\,868.244\,852(20) - 198\,247.689\,374 + 22.234\,505\,83(21) + 198\,244.327\,123 - 195\,865.982\,837 + 198\,242.993\,622 - 195\,274.486\,618 - 330.171\,465 \approx 2991.557\,284$ and $22.688\,312\,0 + 2950.819\,774 + 21.285\,275\,0 \approx 2994.793\,361$ GHz. These two estimates can be associated with the uncertainties of $\sqrt{56^2 + 20^2 + 20^2 + 0.21^2 + 20^2 + 20^2 + 20^2 + 29^2} \approx 80$ and $\sqrt{5^2 + 99^2 + 5^2} \approx 99$ kHz, respectively. These path-based frequencies, together with the MARVEL-predicted values utilizing the full network information [2991.557 31(11) and 2994.793 39(13) GHz, see Table 4], are at least 10 times more accurate than the direct brown links of 83PoMa [54]. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

ments. The correct designations of these lines would be 2405.125, 2991.557, and 2994.793 GHz, respectively. The origin of the incorrect designations can be found in the last column of Table 4. It would be interesting to see if new, ultraprecise measurements confirmed the proposed empirical line positions of this study (see also Fig. 5).

5. Conclusions

One of the aims of the present study has been a comprehensive revision and extension of previous MARVEL analyses [13,22] of the measured rovibrational transitions and energy levels of $^{14}\text{NH}_3$. A significant restriction, making the present study drastically different from all previous MARVEL investigations, is that only those transitions were considered here which involve states with energies less than 7500 cm^{-1} and our study is also constrained to the PCs of the experimental SN of $^{14}\text{NH}_3$. The most notable characteristics and results of the present investigation can be summarized as follows:

(1) The final empirical energy list contains 4936 uniquely labelled states below 7500 cm^{-1} . In a previous similar study [22] the number of empirical energy levels below 7500 cm^{-1} was 4716. These two numbers should be compared to about 26 500, the number of rovibrational energy levels first-principles computations suggest below the energy cutoff of 7500 cm^{-1} .

- (2) While we treated significantly (65%) more transitions and involved 28 new sources during this investigation, compared to Al-Derzi et al. [22], the number of energy levels increased only marginally, by 5%. This points toward the need of an improved design of high-resolution spectroscopic measurements to significantly increase the utility of new data.
- (3) The rovibrational measurements covered in this study involve all possible vibrational bands only up to 4000 cm^{-1} , the first vibrational band missing is at 4008 cm^{-1} . Furthermore, about half of the vibrational states are missing between 4000 and 6000 cm^{-1} . The situation deteriorates quickly above 6000 cm^{-1} . To remedy this problem requires careful experimental studies of the region above 4000 cm^{-1} , which should be helped considerably by the results presented here.
- (4) The empirical rovibrational levels of $^{14}\text{NH}_3$ form a complete set up to 2412.7 cm^{-1} .

In the second part of our study we have identified problems with previous effective Hamiltonian fits associated with floating components leading to correctly fitted transition frequencies but erroneous energy levels. We were able to cure this problem by the judicious use of computed transitions which link the various floating components with the principal components. This issue is a general one for systems which have restrictive selection or propensity rules, such as the dominant $\Delta K = 0$ rule in ammonia. Our procedure, combining network theory and first-principles compu-

tations, offers a way forward in such situations. In fact, using a Watson Hamiltonian with terms up to sixteenth order and utilizing carefully chosen and well-characterized artificial transitions taken from the CoYuTe line list [71] we were able to generate new effective-Hamiltonian parameters for the ground vibrational state of $^{14}\text{NH}_3$, which predict correct energy levels for this state for levels with $K \leq 25$. This effort was, in part, undertaken to address an impasse in the intercomparison of experimental and theoretical energy levels for ammonia. In addition to improving the ability of the effective Hamiltonian to predict energy levels, the method developed here enables the synergetic trust of MARVEL levels that can reliably compare to computed linelists.

Comparisons with our empirical energies help to validate the accuracy of the C2018 PES [70] which underpins the CoYuTe line list. Notwithstanding this comment, the variational nuclear motion calculations used to generate the CoYuTe line list do not produce energy levels with the same accuracy as the MARVEL procedure utilized here. While all the analyses presented in this work which used CoYuTe data used the original computed energy levels, the final CoYuTe line list actually replaced these levels with the MARVEL ones from this work where available [71]. Besides the obvious advantages in terms of improved frequencies, there are a number of specific reasons for doing this. The ExoMol project [72,73], for which CoYuTe was created, aims to produce comprehensive line lists for the characterization of exoplanetary atmospheres. Recently, Doppler-shift high-resolution spectroscopy has been found to be a powerful way of detecting molecules on exoplanets [160]. This technique requires high experimental accuracy for the line positions. Similarly, the most recent HITRAN database [158] is missing a significant number of lines in the 4000 – 7000 cm^{-1} region. The “BARVEL” line list [38] constructed using the previous MARVEL energy levels and the BYTe line list [67] was found to be insufficiently accurate for this purpose [158]; hopefully a line list constructed using the improved CoYuTe intensities and the updated MARVEL energy levels will prove to be more reliable.

Our study confirms the accuracy of a number of measured lines relevant for the efforts of international organizations protecting certain $^{14}\text{NH}_3$ lines under 3 THz [159]. Furthermore, in a few cases we suggest significantly more accurate frequencies for some of the lines, supported by the spectroscopic network of the measured lines.

Finally, the present study was restricted to transition wavenumbers below 7500 cm^{-1} (note that the restriction applied was on the energy levels but this is a direct consequence of our choice). However, ammonia spectra are important at shorter wavelengths than this. For example, visible spectra of ammonia are well known in Jupiter [161]. Recent work has shown that the CoYuTe line list gives the best available model for this spectrum but that there is a significant shift in the line positions [162]. A MARVEL-based study covering these wavelengths would rectify these problems but would require further laboratory high-resolution spectra recorded at near infrared and visible wavelengths first.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

CRediT authorship contribution statement

Tibor Furtenbacher: Methodology, Software, Investigation, Validation, Formal analysis. **Phillip A. Coles:** Data curation, Investigation, Writing - original draft, Writing - review & editing. **Jonathan Tennyson:** Investigation, Supervision, Writing - review & editing, Funding acquisition. **Sergei N. Yurchenko:** Supervision. **Shanshan**

Yu: Software, Validation, Writing - original draft, Writing - review & editing, Formal analysis. **Brian Drouin:** Validation, Writing - original draft, Writing - review & editing, Funding acquisition. **Roland Tóbiás:** Methodology, Software, Investigation, Visualization, Formal analysis. **Attila G. Császár:** Conceptualization, Supervision, Writing - review & editing, Funding acquisition.

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Supplementary material

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.jqsrt.2020.107027.

References

- [1] Galloway JN, Aber JD, Erisman JW, Seitzinger SP, Howarth RW, Cowling EB, et al. The nitrogen cascade. *Bioscience* 2003;53:341–56.
- [2] Erisman JW, Galloway JN, Seitzinger S, Bleeker A, Dise NB, Petrescu AMR, et al. Consequences of human modification of the global nitrogen cycle. *Phil Trans R Soc Lond B* 2013;368:20130116.
- [3] Van Damme M, Clarisse L, Whitburn S, Hadji-Lazaro J, Hurtmans D, Clerbaux C, et al. Industrial and agricultural ammonia point sources exposed. *Nature* 2018;564:99–103.
- [4] Cheung A, Rank DM, Townes CH, Thornton DD, Welch WJ. Detection of NH_3 molecules in the interstellar medium by their microwave emission. *Phys Rev Lett* 1968;21:1701–5.
- [5] Cheung AC, Rank DM, Townes CH, Thornton DD, Welch WJ. Detection of water in interstellar regions by its microwave radiation. *Nature* 1969;221:626–8.
- [6] Ho PTP, Townes CH. Interstellar ammonia. *Ann Rev Astron Astrophys* 1983;21:239–70.
- [7] MacDonald RJ, Madhusudhan N. HD 209458b in new light: evidence of nitrogen chemistry, patchy clouds and sub-solar water. *Mon Not R Astron Soc* 2017;469:1979–96.
- [8] Lucas PW, Tinney CG, Burningham B, Leggett SK, Pinfield DJ, Smart R, et al. The discovery of a very cool, very nearby brown dwarf in the Galactic plane. *Mon Not R Astron Soc* 2010;408:L56–60.
- [9] Schneider AC, Cushing MC, Kirkpatrick JD, Gelino CR, Mace GN, Wright EL, et al. Hubble space telescope spectroscopy of brown dwarfs discovered with the wide-field infrared survey explorer. *Astrophys J* 2015;804:92.
- [10] Canty JI, Lucas PW, Tennyson J, Yurchenko SN, Leggett SK, Tinney CG, et al. Methane and ammonia in the near-infrared spectra of late-T dwarfs. *Mon Not R Astron Soc* 2015;450:454–80.
- [11] Cleeton CE, Williams NH. Electromagnetic waves of 1.1 cm wave-length and the absorption spectrum of ammonia. *Phys Rev* 1934;45:234–7.
- [12] Klopper W, Samson CCM, Tarczay G, Császár AG. Equilibrium inversion barrier of NH_3 from extrapolated coupled-cluster pair energies. *J Comp Chem* 2001;22:1306–14.
- [13] Császár AG, Furtenbacher T. Promoting and inhibiting tunneling via nuclear motions. *Phys Chem Chem Phys* 2016;18:1092–104.
- [14] Fábri C, Marquardt R, Császár AG, Quack M. Controlling tunneling in ammonia isotopomers. *J Chem Phys* 2019;150:014102.
- [15] Markov N. A quantum propulsion method. In: Georgiev P, Soares CG, editors. Sustainable development and innovations in marine technologies; vol. 3 of proceedings in marine technology and ocean engineering. Maritime Assoc Mediterranean; Tech Univ Varna; 2020. p. 247–51. ISBN 978-0-367-81008-5; 978-0-367-40951-7 18th International Congress of the Maritime-Association-of-the-Mediterranean (IMAM), Varna, BULGARIA, SEP 09–11, 2019.
- [16] Lemarchand C, Triki M, Darquie B, Borde CJ, Chardonnet C, Daussey C. Progress towards an accurate determination of the Boltzmann constant by Doppler spectroscopy. *New J Phys* 2011;13:073028.

- [17] Triki M, Lemarchand C, Darquie B, Sow PLT, Roncin V, Chardonnet C, et al. Speed-dependent effects in NH₃ self-broadened spectra: towards the determination of the Boltzmann constant. *Phys Rev A* 2012;85:062510.
- [18] Lemarchand C, Mejri S, Sow PLT, Triki M, Tokunaga SK, Briauudeau S, et al. A revised uncertainty budget for measuring the Boltzmann constant using the Doppler broadening technique on ammonia. *Metrologia* 2013;50:623–30.
- [19] Bethlem HL, Kajita M, Sartakov B, Meijer G, Ubachs W. Prospects for precision measurements on ammonia molecules in a fountain. *Eur Phys J D* 2008;163:55–69.
- [20] Owens A, Yurchenko SN, Thiel W, Špirko V. Accurate prediction of the ammonia probes of a variable proton-to-electron mass ratio. *Mon Not R Astron Soc* 2015;450:3191–200.
- [21] Owens A, Yurchenko SN, Thiel W, Špirko V. Enhanced sensitivity to a possible variation of the proton-to-electron mass ratio in ammonia. *Phys Rev A* 2016;93:052506.
- [22] Al Derzi AR, Furtenbacher T, Yurchenko SN, Tennyson J, Császár AG. MARVEL analysis of the measured high-resolution spectra of ¹⁴NH₃. *J Quant Spectrosc Radiat Transf* 2015;161:117–30.
- [23] Furtenbacher T, Császár AG, Tennyson J. MARVEL: Measured active rotational-vibrational energy levels. *J Mol Spectrosc* 2007;245:115–25.
- [24] Furtenbacher T, Császár AG. MARVEL: measured active rotational-vibrational energy levels. II. Algorithmic improvements. *J Quant Spectrosc Radiat Transf* 2012;113:929–35.
- [25] Tóbiás R, Furtenbacher T, Tennyson J, Császár AG. Accurate empirical rovibrational energies and transitions of H₂¹⁶O. *Phys Chem Chem Phys* 2019;21:3473–95.
- [26] Császár AG, Furtenbacher T. Spectroscopic networks. *J Mol Spectrosc* 2011;266:99–103.
- [27] Furtenbacher T, Császár AG. The role of intensities in determining characteristics of spectroscopic networks. *J Molec Struct (THEOCHEM)* 2012;1009:123–9.
- [28] Császár AG, Furtenbacher T, Árendás P. Small molecules – big data. *J Phys Chem A* 2016;120:8949–69.
- [29] Yachmenev A, Kuepper J. Communication: General variational approach to nuclear-quadrupole coupling in rovibrational spectra of polyatomic molecules. *J Chem Phys* 2017;147:141101.
- [30] Twagirayezu S, Hall GE, Sears TJ. Quadrupole splittings in the near-infrared spectrum of ¹⁴NH₃. *J Chem Phys* 2016;145:144302.
- [31] Aroui H, BenMabrouk K, Boussassi R. Line-mixing effect on NH₃ line intensities. *J Quant Spectrosc Radiat Transf* 2013;130:273–83.
- [32] JBarton E, Yurchenko SN, Tennyson J, Clausen S, Fateev A. High-resolution absorption measurements of NH₃ at high temperatures: 500–2100 cm⁻¹. *J Quant Spectrosc Radiat Transf* 2015;167:126–34.
- [33] Pearson JC, Yu S, Pirali O. Modeling the spectrum of the 2ν₂ and ν₄ states of ammonia to experimental accuracy. *J Chem Phys* 2016;145:124301.
- [34] Sung K, Yu S, Pearson J, Pirali O, Tchana FK, Manceron L. Far-infrared ¹⁴NH₃ line positions and intensities measured with a FT-IR and AILES beamline, synchrotron SOLEIL. *J Mol Spectrosc* 2016;327:1–20.
- [35] Barton EJ, Yurchenko SN, Tennyson J, Béguier S, Campargue A. A near infrared line list for NH₃: analysis of a Kitt Peak spectrum after 35 years. *J Mol Spectrosc* 2016;325:7–12.
- [36] Fusina L, Di Lonardo G, Cane E, Predoi-Cross A, Rozario H, Herman M. The high resolution spectrum of ¹⁵NH₃ in the far-infrared: rotation-inversion transitions in the ground, ν₂ = 1, 2 and ν₄ = 1 states. *J Quant Spectrosc Radiat Transf* 2017;203:417–24.
- [37] Barton EJ, Yurchenko SN, Tennyson J, Clausen S, Fateev A. High-resolution absorption measurements of NH₃ at high temperatures: 2100 – 5500 cm⁻¹. *J Quant Spectrosc Radiat Transf* 2017;189:60–5.
- [38] Beale CA, Hargreaves RJ, Coles P, Tennyson J, Bernath PF. Infrared absorption spectra of hot ammonia. *J Quant Spectrosc Radiat Transf* 2017;203:410–16.
- [39] Barton EJ, Polyansky OL, Yurchenko SN, Tennyson J, Civis S, Ferus M, et al. Absorption spectra of ammonia near 1 μm. *J Quant Spectrosc Radiat Transf* 2017;203:392–7.
- [40] Zobov NE, Coles PA, Ovsyannikov RI, Kyuberis AA, Hargreaves RJ, Bernath PF, et al. Analysis of the red and green optical absorption spectrum of gas phase ammonia. *J Quant Spectrosc Radiat Transf* 2018;209:224–31.
- [41] Coles PA, Ovsyannikov RI, Polyansky OL, Yurchenko SN, Tennyson J. Improved potential energy surface and spectral assignments for ammonia in the near-infrared region. *J Quant Spectrosc Radiat Transf* 2018;219:199–212.
- [42] Kukolich SG. Measurement of hyperfine structure of the J = 3, K = 2 inversion line of ¹⁴NH₃. *Phys Rev* 1965;138(5A):A1322–5.
- [43] Kukolich SG. Measurement of ammonia hyperfine structure with a two-cavity maser. *Phys Rev* 1967;156:83–92.
- [44] Kukolich SG, Wofsy SC. ¹⁴NH₃ hyperfine structure and quadrupole coupling. *J Chem Phys* 1970;52:5477–81.
- [45] Freund SM, Oka T. Infrared-microwave two-photon spectroscopy: the ν₂ band of NH₃. *Phys Rev A* 1976;13:2178–90.
- [46] Hillman JJ, Kostiuk T, Buhl D, Faris JL, Novaco JC, Mumma MJ. Precision measurements of NH₃ spectral lines near 11 μm using the infrared heterodyne technique. *Opt Lett* 1977;1(3):81.
- [47] Kostiuk T, Mumma MJ, Hillman JJ, Buhl D, Brown LW, Faris JL. NH₃ spectral line measurements on Earth and Jupiter using 10 mm superheterodyne receiver. *Infrared Phys* 1977;17:431–9.
- [48] Jones H. Infrared-microwave two-photon spectroscopy with ¹³C¹⁶O₂ and ¹²C¹⁸O₂ lasers of the ν₂-band of ammonia. *Appl Phys* 1978;15(3):261–4.
- [49] Belov SP, Gershstein LI, Krupnov AF, Maslovskij AV, Urban Š, Špirko V, et al. Inversion and inversion-rotation spectrum of ¹⁴NH₃ in the ν₂ excited state. *J Mol Spectrosc* 1980;84:288–304.
- [50] Sinha BV, Smith PDP. New microwave inversion lines of ¹⁴NH₃ in the ground state. *J Mol Spectrosc* 1980;80:231–2.
- [51] Sattler JP, Worchesky TL. Additional diode laser heterodyne measurements on ammonia. *J Mol Spectrosc* 1981;90:297–301.
- [52] Minguzzi P, Tonelli M, Carrozzi A, Lieto AD. Optoacoustic laser Stark spectroscopy in the ν₂ band of ¹⁴NH₃. *J Mol Spectrosc* 1982;96(2):294–305.
- [53] Shoja-Chaghervand P, Bjarnov E, Schwendeman R. Infrared-microwave two-photon spectroscopy of the ν₂ band of ¹⁴NH₃. *J Mol Spectrosc* 1983;97(2):287–305.
- [54] Poynter RL, Margolis JS. The ground state far infrared spectrum of NH₃. *Mol Phys* 1983;48:401–18.
- [55] Magerl G, Schupita W, Frye JM, Kreiner WA, Oka T. Sub-doppler spectroscopy of the ν₂ band of NH₃ using microwave modulation sidebands of CO₂ laser lines. *J Mol Spectrosc* 1984;107(1):72–83.
- [56] Siemsen KJ, Reid J. Heterodyne frequency measurements of ¹⁴NH₃ and ¹⁵NH₃ ν₂-band transitions. *Optics Lett* 1985;10:594–6.
- [57] D’Cunha R. The a₂ν₂ ← sν₂ bands of ¹⁴NH₃ and ¹⁵NH₃. *J Mol Spectrosc* 1987;122:130–4.
- [58] Lellouch E, Lacombe N, Guelachvili G, Tarrago G, Encrenaz T. Ammonia: experimental absolute line strengths and self-broadening parameters in the 1800 to 2100 cm⁻¹ range. *J Mol Spectrosc* 1987;124:333–47.
- [59] Snels M, Baldacchini G. Shape and width of IR absorption lines of ammonia expanded in a supersonic jet. *Appl Phys B* 1988;47:277–82.
- [60] Smith PDP, Firth S, Davis RW. High J inversion lines of ammonia. *J Mol Spectrosc* 1990;144:448–50.
- [61] Fabian M, Ito F, Yamada KMT. N₂, O₂, and air broadening of NH₃ in ν₂ band measured by FTIR spectroscopy. *J Mol Spectrosc* 1995;173:591–602.
- [62] Winnewisser G, Belov SP, Klaus T, Urban S. Ro-inversion spectrum of ammonia. *Z Naturforsch* 1996;51:200–6.
- [63] Sun Z-D, Lees RM, Xu L-H. Lamb-dip measurements of ammonia calibration lines in methylamine using a dual-mode CO₂-laser-microwave-sideband spectrometer. *J Mol Spectrosc* 2008;249:68–70.
- [64] Cazzoli G, Dore L, Puzzarini C. The hyperfine structure of the inversion-rotation transition J_K = 1₀ ← 0₀ of NH₃ investigated by Lamb-dip spectroscopy. *Astron Astrophys* 2009;507:1707–10.
- [65] Coy SL, Lehmann KK. Rotational structure of ammonia N-H stretch overtones: five and six quanta bands. *J Chem Phys* 1986;84:5239–49.
- [66] Yurchenko SN, Barber RJ, Yachmenev A, Thiel W, Jensen P, Tennyson J. A variationally computed T = 300 K line list for NH₃. *J Phys Chem A* 2009;113:11845–55.
- [67] Yurchenko SN, Barber RJ, Tennyson J. A variationally computed hot line list for NH₃. *Mon Not R Astron Soc* 2011;413:1828–34.
- [68] Huang X, Schwenke DW, Lee TJ. Rovibrational spectra of ammonia. II. Detailed analysis, comparison, and prediction of spectroscopic assignments for ¹⁴NH₃, ¹⁵NH₃, and ¹⁴NH₃. *J Chem Phys* 2011;134:044321.
- [69] Yurchenko SN. A theoretical room-temperature line list for ¹⁵NH₃. *J Quant Spectrosc Radiat Transf* 2015;152:28–36.
- [70] Coles PA, Ovsyannikov RI, Polyansky OL, Yurchenko SN, Tennyson J. Improved potential energy surface and spectral assignments for ammonia in the near-infrared region. *J Quant Spectrosc Radiat Transf* 2018;219:199–212.
- [71] Coles PA, Yurchenko SN, Tennyson J. ExoMol molecular line lists XXXV: a rotation-vibration line list for hot ammonia. *Mon Not R Astron Soc* 2019;490:4638–47.
- [72] Tennyson J, Yurchenko SN. ExoMol: molecular line lists for exoplanet and other atmospheres. *Mon Not R Astron Soc* 2012;425:21–33.
- [73] Tennyson J, Yurchenko SN, Al-Refaie AF, Barton EJ, Chubb KL, Coles PA, et al. The ExoMol database: molecular line lists for exoplanet and other hot atmospheres. *J Mol Spectrosc* 2016;327:73–94.
- [74] Costain CC. An empirical formula for the microwave spectrum of ammonia. *Phys Rev* 1951;82:108.
- [75] Sasada H, Endo Y, Hirota E, Poynter RL, Margolis JS. Microwave and Fourier-transform infrared spectroscopy of the ν₄ = 1 and ν₂ = 2 states of NH₃. *J Mol Spectrosc* 1992;151:33–53.
- [76] Kleiner I, Tarrago G, Brown LR. Positions and intensities in the 3ν₂/ν₂ + ν₄ vibrational system of ¹⁴NH₃ near 4 μm. *J Mol Spectrosc* 1995;173:120–45.
- [77] Cottaz C, Kleiner I, Tarrago G, Brown LR, Margolis JS, Poynter RL, et al. Line positions and intensities in the 2ν₂/ν₄ vibrational system of ¹⁴NH₃ near 5–7 μm. *J Mol Spectrosc* 2000;203:285–309.
- [78] Cottaz C, Tarrago G, Kleiner I, Brown LR. Assignments and intensities of ¹⁴NH₃ hot bands in the 5- to 8-μm (3ν₂ - ν₂, ν₂ + ν₄ - ν₂) and 4-μm (4ν₂ - ν₂, ν₁ - ν₂, ν₃ - ν₂ and 2ν₄ - ν₂) regions. *J Mol Spectrosc* 2001;209:30–49.
- [79] Sung K, Brown LR, Huang X, Schwenke DW, Lee TJ, Coy SL, et al. Extended line positions, intensities, empirical lower state energies and quantum assignments of NH₃ from 6300 to 7000 cm⁻¹. *J Quant Spectrosc Radiat Transf* 2012;113:1066–83.
- [80] Chen P, Pearson JC, Pickett HM, Matsuura S, Blake GA. Measurements of ¹⁴NH₃ in the ν₂ = 1 state by a solid-state, photomixing, THz spectrometer, and a simultaneous analysis of the microwave, terahertz, and infrared transitions between the ground and ν₂ inversion-rotation levels. *J Mol Spectrosc* 2006;236:116–26.

- [81] Yu S, Pearson JC, Drouin BJ, Sung K, Pirali O, Vervloet M, et al. Submillimeter-wave and far-infrared spectroscopy of high- J transitions of the ground and $v_2 = 1$ states of ammonia. *J Chem Phys* 2010;133:174317.
- [82] Pearson J, Yu S, Pearson J, Sung K, Drouin B, Pirali O. Extended measurements and an experimental accuracy effective hamiltonian model for the $3\nu_2$ and $\nu_4 + \nu_2$ states of ammonia. *J Mol Spectrosc* 2018;353:60–6.
- [83] Furtenbacher T, Császár AG. MARVEL: measured active rotational-vibrational energy levels. II. Algorithmic improvements. *J Quant Spectrosc Radiat Transf* 2012;113:929–35.
- [84] Árendás P, Furtenbacher T, Császár AG. On spectra of spectra. *J Math Chem* 2016;54:806–22.
- [85] Császár AG, Czákó G, Furtenbacher T, Mátyus E. An active database approach to complete rotational-vibrational spectra of small molecules. *Annu Rep Comput Chem* 2007;3:155–76.
- [86] Furtenbacher T, Árendás P, Mellau G, Császár AG. Simple molecules as complex systems. *Sci Rep* 2014;4:4654.
- [87] Furtenbacher T, Szabó I, Császár AG, Bernath PF, Yurchenko SN, Tennyson J. Experimental energy levels and partition function of the $^{12}\text{C}_2$ molecule. *Astrophys J Suppl* 2016;224:44.
- [88] McKemmish LK, Masseron T, Sheppard S, Sandeman E, Schofield Z, Furtenbacher T, et al. MARVEL analysis of the measured high-resolution spectra of $^{48}\text{Tl}^{16}\text{O}$. *Astrophys J Suppl* 2017;228:15.
- [89] McKemmish LK, Borsosvsky J, Goodhew KL, Sheppard S, Bennett AFV, Martin ADJ, et al. MARVEL analysis of the measured high-resolution spectra of $^{90}\text{Zr}^{16}\text{O}$. *Astrophys J* 2018;867:33.
- [90] Darby-Lewis D, Shah H, Joshi D, Kahn F, Kauwo M, Sethi N, et al. MARVEL analysis of the measured high-resolution spectra of ^{14}NH . *J Mol Spectrosc* 2019;362:69–76.
- [91] Furtenbacher T, Horváth M, Koller D, Sólyom P, Balogh A, Balogh I, et al. MARVEL analysis of the measured high-resolution rovibronic spectra and definitive ideal-gas thermochemistry of the $^{16}\text{O}_2$ molecule. *J Phys Chem Ref Data* 2019;48:023101.
- [92] Furtenbacher T, Szidarovszky T, Mátyus E, Fábri C, Császár AG. Analysis of the rotational-vibrational states of the molecular ion H_3^+ . *J Chem Theory Comput* 2013;9:5471–8.
- [93] Furtenbacher T, Szidarovszky T, Fábri C, Császár AG. MARVEL analysis of the rotational-vibrational states of the molecular ions H_2D^+ and D_2H^+ . *Phys Chem Chem Phys* 2013;15:10181–93.
- [94] Tennyson J, Bernath PF, Brown LR, Campargue A, Carleer MR, Császár AG, et al. IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part I. Energy levels and transition wavenumbers for H_2^{17}O and H_2^{18}O . *J Quant Spectrosc Radiat Transf* 2009;110:573–96.
- [95] Tennyson J, Bernath PF, Brown LR, Campargue A, Carleer MR, Császár AG, et al. IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part II. Energy levels and transition wavenumbers for HD^{16}O , HD^{17}O , and HD^{18}O . *J Quant Spectrosc Radiat Transf* 2010;111:2160–84.
- [96] Tennyson J, Bernath PF, Brown LR, Campargue A, Carleer MR, Császár AG, et al. IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part III. Energy levels and transition wavenumbers for H_2^{16}O . *J Quant Spectrosc Radiat Transf* 2013;117:29–80.
- [97] Tennyson J, Bernath PF, Brown LR, Campargue A, Császár AG, Daumont L, et al. IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part IV. Energy levels and transition wavenumbers for D_2^{16}O , D_2^{17}O and D_2^{18}O . *J Quant Spectrosc Radiat Transf* 2014;142:93–108.
- [98] Tennyson J, Bernath PF, Brown LR, Campargue A, Császár AG, Daumont L, et al. A database of water transitions from experiment and theory (IUPAC Technical Report). *Pure Appl Chem* 2014;86:71–83.
- [99] Tóbiás R, Furtenbacher T, Császár AG, Naumenko OV, Tennyson J, Flaud J-M, et al. Critical evaluation of measured rotational-vibrational transitions of four sulphur isotopologues of S^{16}O_2 . *J Quant Spectrosc Radiat Transf* 2018;208:152–63.
- [100] Chubb KL, Naumenko OV, Keely S, Bartolotto S, MacDonald S, Mukhtar M, et al. MARVEL analysis of the measured high-resolution rovibrational spectra of H_2S . *J Quant Spectrosc Radiat Transf* 2018;218:178–86.
- [101] Wang Y, Owens A, Tennyson J, Yurchenko SN. MARVEL analysis of the measured high-resolution rovibronic spectra of the calcium monohydroxide radical (CaOH). *Astrophys J Suppl* 2020.
- [102] Chubb KL, Joseph M, Franklin J, Choudhury N, Furtenbacher T, Császár AG, et al. MARVEL analysis of the measured high-resolution spectra of C_2H_2 . *J Quant Spectrosc Radiat Transf* 2018;204:42–55.
- [103] Fábri C, Mátyus E, Furtenbacher T, Nemes L, Mihály B, Zoltáni T, et al. Variational quantum mechanical and active database approaches to the rotational-vibrational spectroscopy of ketene, H_2CCO . *J Chem Phys* 2011;135:094307.
- [104] Bunker PR, Jensen P, editors. *Molecular symmetry and spectroscopy*, 2nd ed.. NRC Research Press, Ottawa, Ontario, Canada; 1998.
- [105] Urban Š, D’Cunha R, Narahari Rao K, Papoušek D. The $\Delta k = \pm 2$ “forbidden band” and inversion-rotation energy levels of ammonia. *Can J Phys* 1984;62:1775–91.
- [106] Tóbiás R, Furtenbacher T, Simkó I, Császár AG, Diouf ML, Cozijn FMJ, et al. Spectroscopic-network-assisted precision spectroscopy and its application to water. *Nature Comm* 2020;11:1708.
- [107] Down MJ, Hill C, Yurchenko SN, Tennyson J, Brown LR, Kleiner I. Re-analysis of ammonia spectra: updating the HITRAN $^{14}\text{NH}_3$ database. *J Quant Spectrosc Radiat Transf* 2013;130:260–72.
- [108] Huang X.. Xinchuan Huang SETI webpage; 2013. <http://huang.seti.org/>.
- [109] Cohen EA, Poynter RL. The microwave spectrum of $^{14}\text{NH}_3$ in the $\nu = 000^011$ state. *J Mol Spectrosc* 1974;53:131–9.
- [110] Poynter RL, Kakar RK. The microwave frequencies, line parameters, and spectral constants for $^{14}\text{NH}_3$. *Astrophys J Suppl* 1975;29:87–96.
- [111] Tanaka K, Endo Y, Hirota E. Submillimeter-wave spectrum of $k = \pm 1 \leftarrow \mp 2$ transitions of NH_3 . *Chem Phys Lett* 1988;146:165–8.
- [112] Sasada H, Hasegawa Y, Amano T, Shimizu T. High-resolution infrared and microwave spectroscopy of the ν_4 and $2\nu_2$ bands of $^{14}\text{NH}_3$ and $^{15}\text{NH}_3$. *J Mol Spectrosc* 1982;96:106–30.
- [113] Fichoux H, Khelkhal M, Rusinek E, Legrand J, Herlemont F, Urban Š. Double resonance sub-Doppler study of the allowed and $\Delta K = -3$ forbidden $Q(3,3)$ transitions to the ν_2 vibrational state of $^{14}\text{NH}_3$. *J Mol Spectrosc* 1998;192:169–78.
- [114] Cohen EA. The ν_4 state inversion spectra of $^{15}\text{NH}_3$ and $^{14}\text{NH}_3$. *J Mol Spectrosc* 1980;79:496–501.
- [115] Drouin BJ, Yu S, Pearson JC, Gupta H. Terahertz spectroscopy for space applications: 2.5–2.7 THz spectra of HD, H_2O and NH_3 . *J Mol Struct* 2011;1006:2–12.
- [116] Krupnov AF, Tretyakov MY, Bogey M, Bailleux S, Walters A, Delcroix B, et al. Microwave measurements of $J = 2 \leftarrow 1, K = 0, 1$ ammonia transitions at 1.215 THz. *J Mol Spectrosc* 1996;176:442–3.
- [117] Belov SP, Urban Š, Winniewisser G. Hyperfine structure of rotation-inversion levels in the excited ν_2 state of ammonia. *J Mol Spectrosc* 1998;189:1–7.
- [118] Urban Š, Herlemont F, Khelkhal M, Fichoux H, Legrand J. High-accuracy determination of the frequency and quadrupole structure of the $sP(1,0)$ and $aR(0,0)$ transitions to the ν_2 state of $^{14}\text{NH}_3$. *J Mol Spectrosc* 2000;200:280–2.
- [119] Guinet M, Jeseck P, Mondelain D, Pepin I, Janssen C, Camy-Peyret C, et al. Absolute measurements of intensities, positions and self-broadening coefficients of R branch transitions in the ν_2 band of ammonia. *J Quant Spectrosc Radiat Transf* 2011;112:1950–60.
- [120] Brown LR, Toth RA. Comparison of the frequencies of NH_3 , CO_2 , H_2O , N_2O , CO , and CH_4 as infrared calibration standards. *J Opt Soc Am B* 1985;2:842–56.
- [121] Sasada H, Schwendeman RH, Magerl G, Poynter RL, Margolis JS. High-resolution spectroscopy of the $\nu_2 = 2a \leftarrow \nu_2 = 1s$ band of $^{14}\text{NH}_3$. *J Mol Spectrosc* 1986;117:317–30.
- [122] Urban Š, Tu N, Narahari Rao K, Guelachvili G. Analysis of high-resolution Fourier transform spectra of $^{14}\text{NH}_3$ at 2.3 μm . *J Mol Spectrosc* 1989;133:312–30.
- [123] Sattler JP, Miller LS, Worchesky TL. Diode laser heterodyne measurements on $^{14}\text{NH}_3$. *J Mol Spectrosc* 1981;88:347–51.
- [124] Urban Š, Papoušek D, Kauppinen J, Yamada K, Winniewisser G. The ν_2 band of $^{14}\text{NH}_3$: a calibration standard with better than $1 \times 10^{-4} \text{ cm}^{-1}$ precision. *J Mol Spectrosc* 1983;101:1–15.
- [125] Poynter RL, Margolis JS. The ν_2 spectrum of NH_3 . *Mol Phys* 1984;51:393–412.
- [126] Chu Z, Chen L, Cheo PK. Absorption spectra of NH_3 using a microwave-sideband CO $_2$ -laser spectrometer. *J Quant Spectrosc Radiat Transf* 1994;51:591–602.
- [127] Hillman JJ, Jennings DE, Faris JL. Diode laser-CO $_2$ laser heterodyne spectrometer: measurement of $2sQ(1, 1)$ in $2\nu_2 - \nu_2$ of NH_3 . *Appl Opt* 1979;18:1808–11.
- [128] Urban Š, Špirko V, Papoušek D, Kauppinen J, Belov SP, Gershtein LI, et al. A simultaneous analysis of the microwave, submillimeterwave, far infrared, and infrared-microwave two-photon transitions between the ground and ν_2 inversion-rotation levels of $^{14}\text{NH}_3$. *J Mol Spectrosc* 1981;88:274–92.
- [129] Weber WH. Laser-Stark spectrum of NH_3 with the CO laser: determination of the ground state dipole moment. *J Mol Spectrosc* 1984;107:405–16.
- [130] Guelachvili G, Abdullah AH, Tu N, Narahari Rao K, Urban Š, Papoušek D. Analysis of high-resolution Fourier-transform spectra of $^{14}\text{NH}_3$ at 3.0 μm . *J Mol Spectrosc* 1989;133:345–64.
- [131] Pine AS, Dang-Nhu M. Spectral intensities in the ν_1 band of NH_3 . *J Quant Spectrosc Radiat Transf* 1993;50:565–70.
- [132] Brown LR, Margolis JS. Empirical line parameters of NH_3 from 4791 to 5294 cm^{-1} . *J Quant Spectrosc Radiat Transf* 1996;56:283–94.
- [133] Fabian M, Yamada KMT. Absolute intensity of the $\text{NH}_3\nu_2$ band. *J Mol Spectrosc* 1999;198:102–9.
- [134] Kleiner I, Brown LR, Tarrago G, Kou Q-L, Picqué N, Guelachvili G, et al. Positions and intensities in the $2\nu_4/\nu_1/\nu_3$ vibrational system of $^{14}\text{NH}_3$ near 3 μm . *J Quant Spectrosc Radiat Transf* 1999;193:46–71.
- [135] Földes T, Golebiowski D, Herman M, Softley TP, Di Lonardo G, Fusina L. Low-temperature high-resolution absorption spectrum of $^{14}\text{NH}_3$ in the $\nu_1 + \nu_3$ band region (1.51 μm). *Mol Phys* 2014;112:2407–18.
- [136] Urban Š, D’Cunha R, Manheim J, Narahari Rao K. High- J transitions in the ν_2 bands of $^{14}\text{NH}_3$ and $^{15}\text{NH}_3$. *J Mol Spectrosc* 1986;118:298–309.
- [137] Nereson N. Diode laser measurements of NH_3 absorption lines around 10.6 μm . *J Mol Spectrosc* 1978;69:489–93.
- [138] Papoušek D, Urban S, Spirko V, Rao KN. The $\Delta k = \pm 2$ and $\Delta k = \pm 3$ forbidden transitions in the vibrational-rotational spectra of symmetric top molecules NH_3 and H_3O^+ . *J Mol Spectrosc* 1986;141:316–6.
- [139] Brown LR, Peterson DB. An empirical expression for linewidths of ammonia from far-infrared measurements. *J Mol Spectrosc* 1994;168:593–606.
- [140] Angström R, Finsterholz H, Frunder H, Illig D, Papoušek D, Pračna P, et al. Fourier transform and CARs spectroscopy of the ν_1 and ν_3 fundamental bands of $^{14}\text{NH}_3$. *J Mol Spectrosc* 1985;114:454–72.
- [141] Zobov NF, Shirin SV, Ovsyannikov RI, Polyansky OL, Yurchenko SN, Barber RJ, et al. Analysis of high temperature ammonia spectra from 780 to 2100 cm^{-1} . *J Mol Spectrosc* 2011;269:104–8.

- [142] Down MJ, Hill C, Yurchenko SN, Tennyson J, Brown LR, Kleiner I. Re-analysis of ammonia spectra: updating the HITRAN $^{14}\text{NH}_3$ database. *J Quant Spectrosc Radiat Transf* 2013;130:260–72.
- [143] Cermák P, Hovorka J, Veis P, Cacciani P, Cosléou J, Romh JE, et al. Spectroscopy of $^{14}\text{NH}_3$ and $^{15}\text{NH}_3$ in the 2.3 μm spectral range with a new VECSEL laser source. *J Quant Spectrosc Radiat Transf* 2014;137:13–22.
- [144] Hermanussen J, Bizzarri A, Baldacchini G. Diode laser measurements of ammonia absorption lines over the range 620–740 cm^{-1} . *J Mol Spectrosc* 1986;119:291–8.
- [145] Li L, Lees RM, Xu L-H. External cavity tunable diode laser spectra of the $\nu_1 + 2\nu_4$ stretch-band combination bands of $^{14}\text{NH}_3$ and $^{15}\text{NH}_3$. *J Mol Spectrosc* 2007;243:219–26.
- [146] Lees RM, Li L, Xu L-H. New VISTA on ammonia in the 1.5 μm region: assignments for the $\nu_3 + 2\nu_4$ bands of $^{14}\text{NH}_3$ and $^{15}\text{NH}_3$ by isotopic shift labeling. *J Mol Spectrosc* 2008;251:241–51.
- [147] Dietiker P, Milogyadov E, Quack M, Schneider A, Seyfang G. Two photon IR-laser induced population transfer in NH_3 – First steps to measure parity violation in chiral molecules. In: Stock D, Wester R, Scheier P, editors. Proceedings of the 19th symposium on atomic, cluster and surface physics 2014 (SASP 2014). Innsbruck University Press, Innsbruck; 2014. p. p..
- [148] Urban Š, Špirko V, Papoušek D, McDowell RS, Nereson NG, Belov SP, et al. Coriolis and *l*-type interactions in the ν_2 , $2\nu_2$, and ν_4 states of $^{14}\text{NH}_3$. *J Mol Spectrosc* 1980;79:455–95.
- [149] Helminger P, De Lucia FC, Gordy W. Rotational spectra of NH_3 and ND_3 in the 0.5-mm wavelength region. *J Mol Spectrosc* 1971;39:94–7.
- [150] Urban Š, Misra P, Narahari Rao K. The $\nu_1 + \nu_2$ and $\nu_1 + \nu_2 - \nu_2$ bands of $^{14}\text{NH}_3$ and $^{15}\text{NH}_3$. *J Mol Spectrosc* 1985;114:377–94.
- [151] Lundsberg-Nielsen L, Hegelund F, Nicolaisen FM. Analysis of the high-resolution spectrum of ammonia ($^{14}\text{NH}_3$) in the near-infrared region, 6400–6900 cm^{-1} . *J Mol Spectrosc* 1993;162:230–45.
- [152] Petersen JC, Hald J. Microwave optical double resonance spectroscopy of ammonia in a hollow-core fiber. *Opt Express* 2010;18:7955–64.
- [153] Berden G, Peeters R, Meijer G. Cavity-enhanced absorption spectroscopy of the 1.5 μm band system of jet-cooled ammonia. *Chem Phys Lett* 1999;307:131–8.
- [154] Xu L-H, Liu Z, Yakovlev I, Tretyakov MY, Lees RM. External cavity tunable diode laser NH_3 spectra in the 1.5 μm region. *Infrared Phys Techn* 2004;45:31–45.
- [155] Barton EJ, Yurchenko SN, Tennyson J, Clausen S, Fateev A. High-resolution absorption measurements of NH_3 at high temperatures: 500 – 2100 cm^{-1} . *J Quant Spectrosc Radiat Transf* 2015;167:126–34.
- [156] Beale CA, Wong A, Bernath P. Infrared transmission spectra of hot ammonia in the 4800–9000 cm^{-1} region. *J Quant Spectrosc Radiat Transf* 2020;246:106911.
- [157] Polyansky OL, Ovsyannikov RI, Kyuberis AA, Lodi L, Tennyson J, Yachmenev A, et al. Calculation of rotation-vibration energy levels of the ammonia molecule based on an *ab initio* potential energy surface. *J Mol Spectrosc* 2016;327:21–30.
- [158] Gordon IE, et al. The HITRAN 2016 molecular spectroscopic database. *J Quant Spectrosc Radiat Transf* 2017;203:3–69.
- [159] National Academies of Sciences Engineering, and Medicine. Frequency Allocations and Spectrum Protection for Scientific Uses, 2nd edition. The National Academies Press, Washington D.C.; 2015.
- [160] Birkby JL. Spectroscopic direct detection of exoplanets. *Handbook of Exoplanets* 2018:1485–508.
- [161] Lutz BL, Owen T. Visible bands of ammonia - band strengths, curves of growth, and the spatial-distribution of ammonia on Jupiter. *Astrophys J* 1980;235:285–93.
- [162] Irwin PGJ, Bowles N, Braude AS, Garland R, Calcutt S, Coles PA, et al. Analysis of gaseous ammonia (NH_3) absorption in the visible spectrum of Jupiter - Update. *Icarus* 2019;321:572–82.